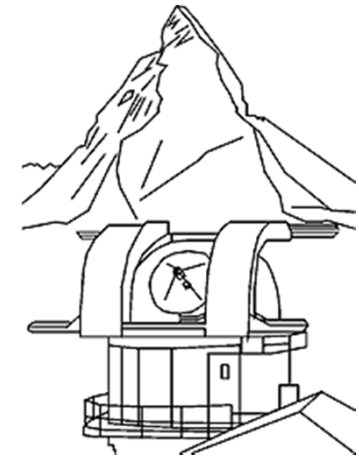


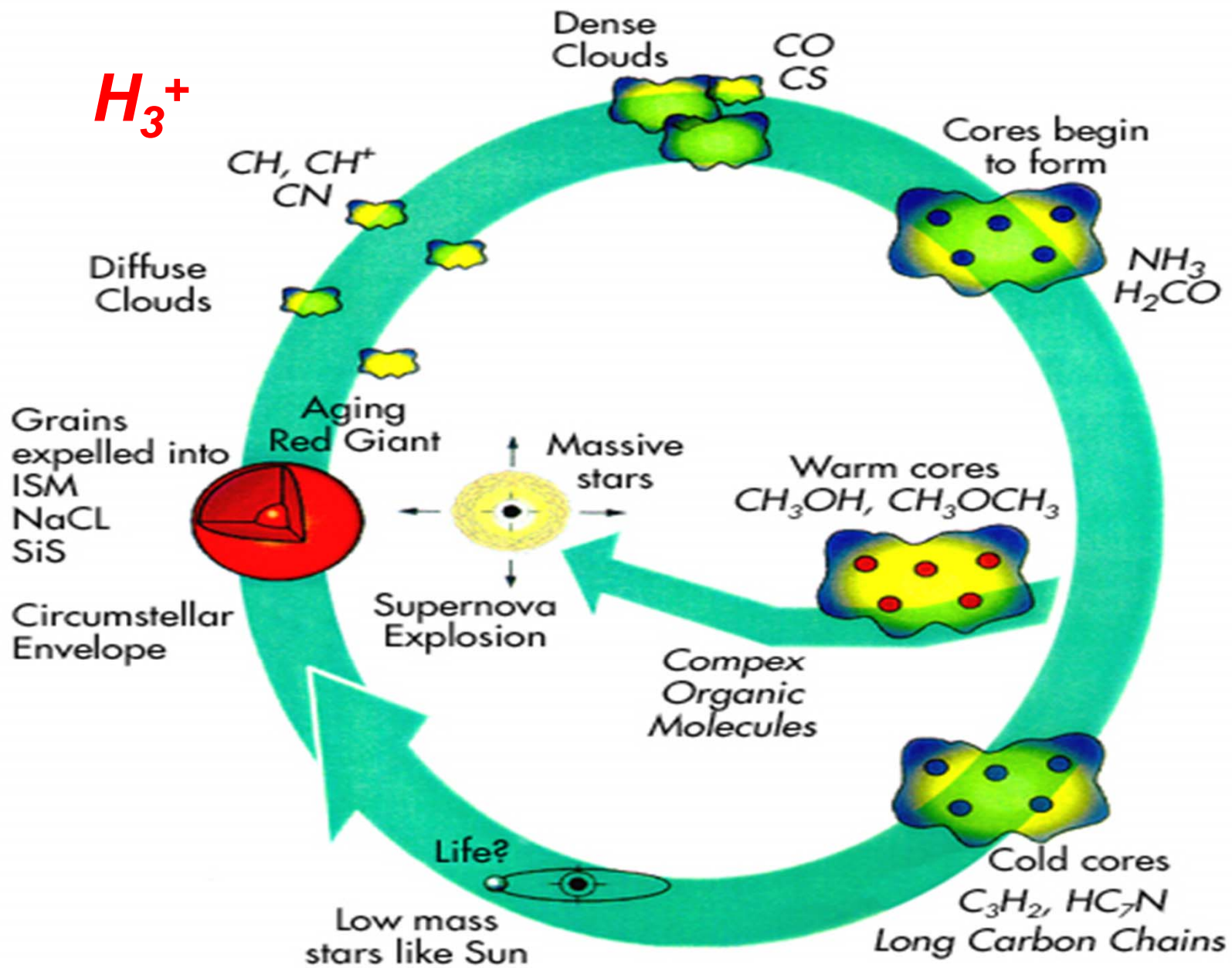
# Cold trap experiments on $\text{H}_3^+ + \text{O}_2$ proton transfer Stephan Schlemmer

Universität zu Köln

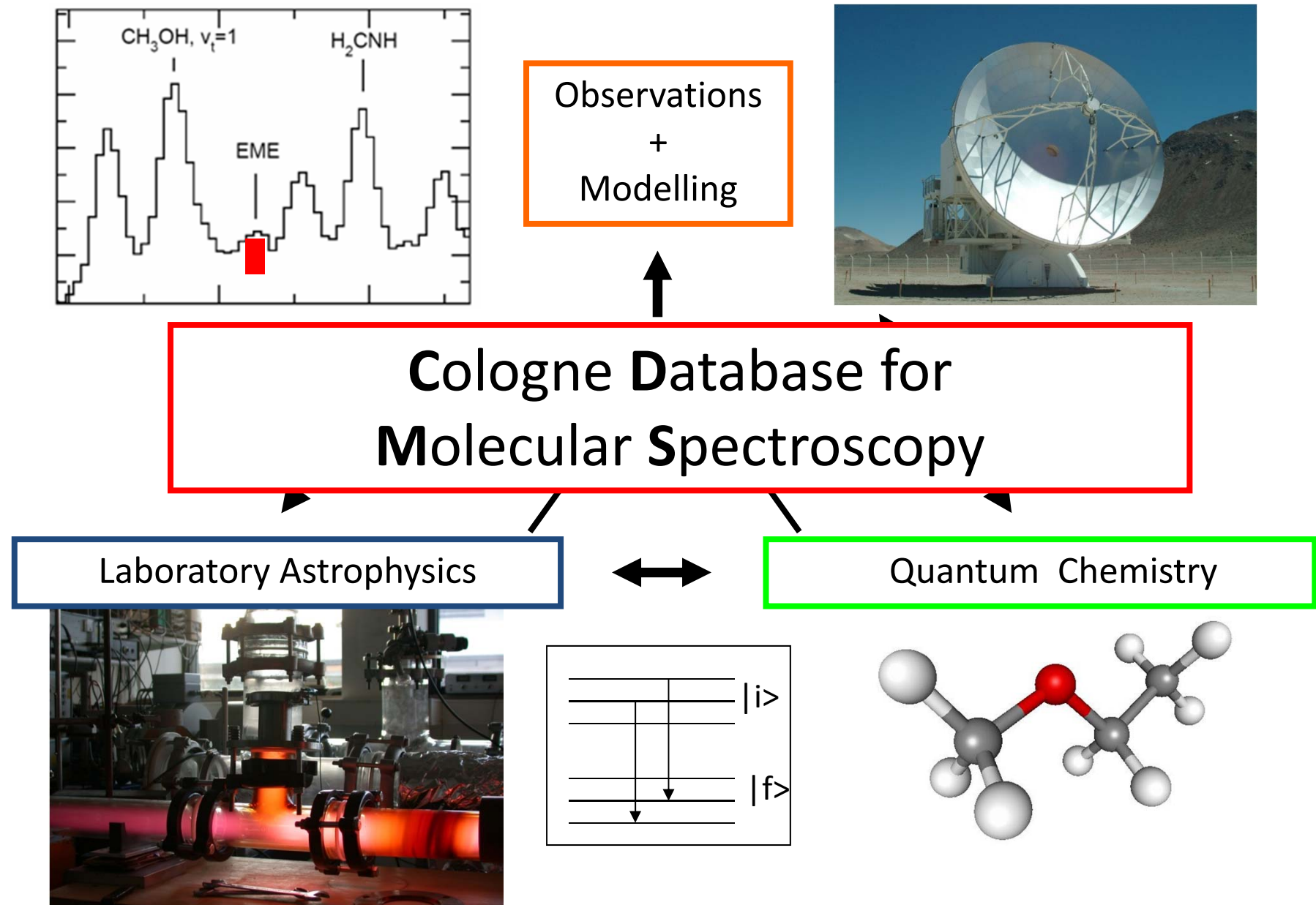


- The THz view into the Sky: Astrophysical Picture
- $\text{H}_2\text{D}^+ / \text{H}_3^+$  in Space and Laboratory
- $\text{H}_3^+ + \text{O}_2$  Proton Transfer
- Preparing a single rotational state of  $\text{H}_3^+$

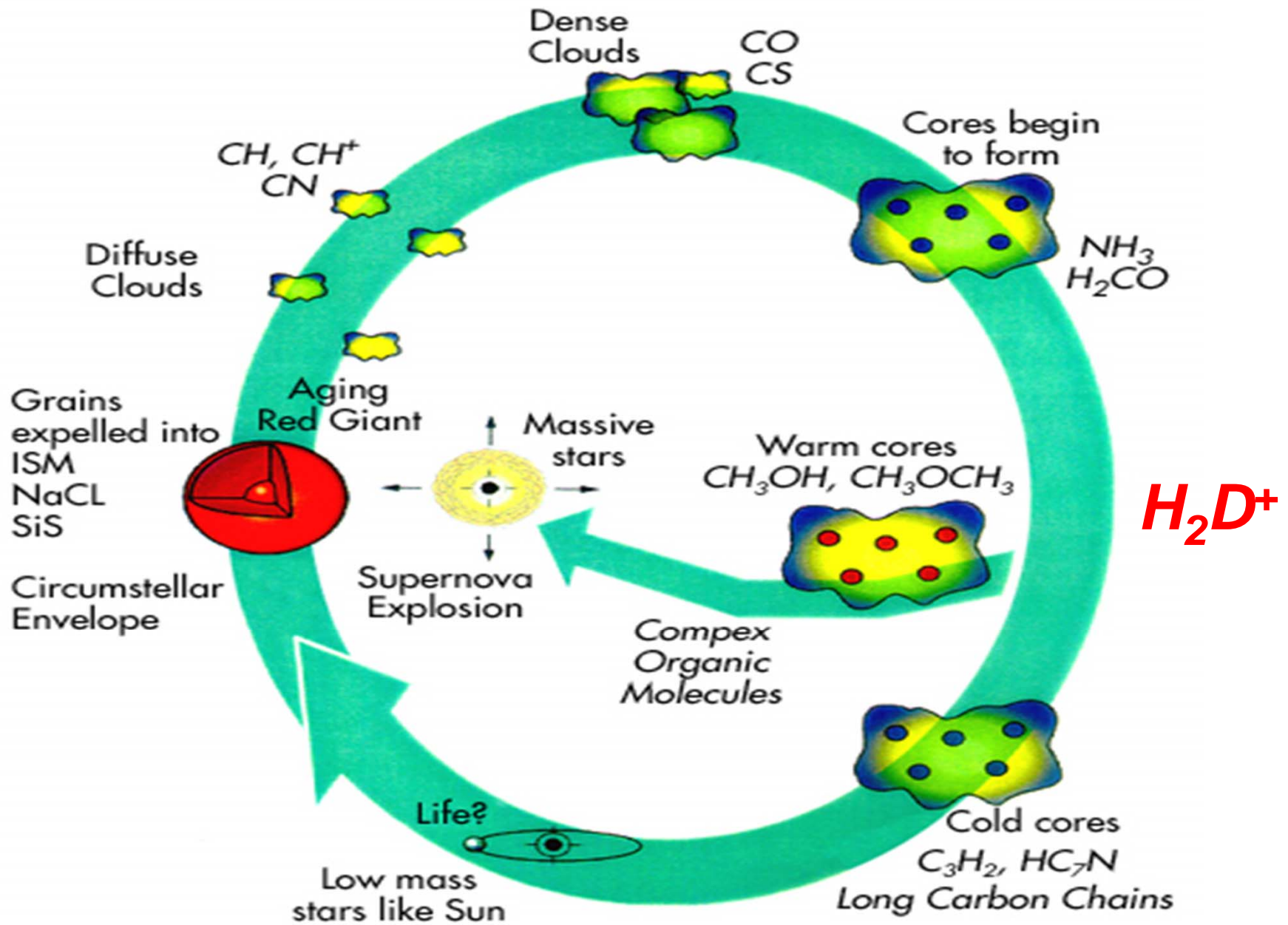
# Life cycle of Stars



# Understanding the Language of Interstellar Molecules



# Molecules: Chemical Clocks



# Deuterated Molecules in Interstellar Medium

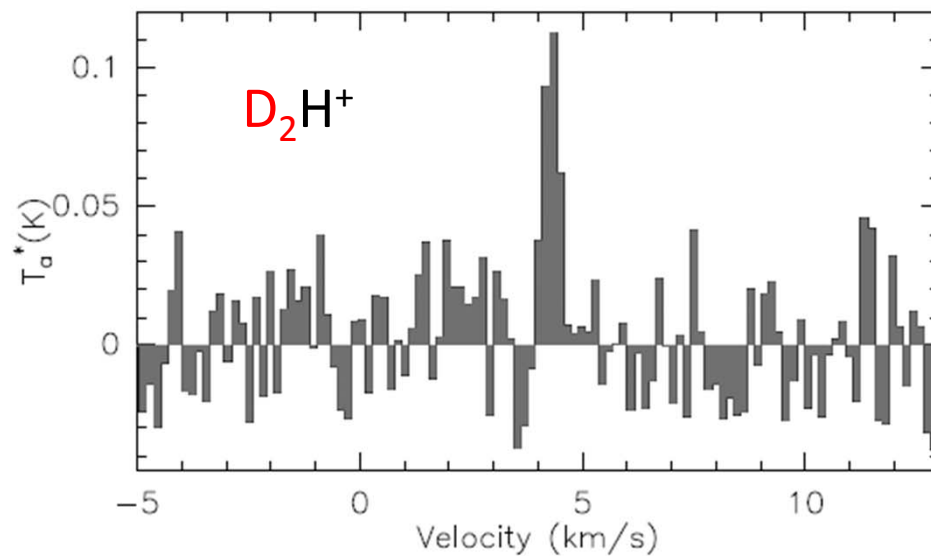
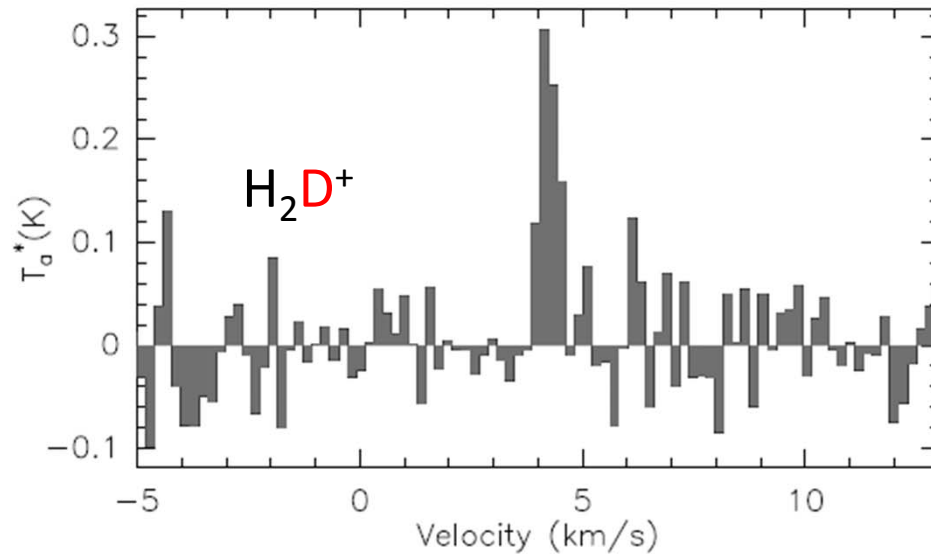
Cosmic  $[D]/[H] \sim 1.5 \cdot 10^{-5}$

Deuteriumreservoir

$[HD]/[H_2] \sim 3.0 \cdot 10^{-5}$

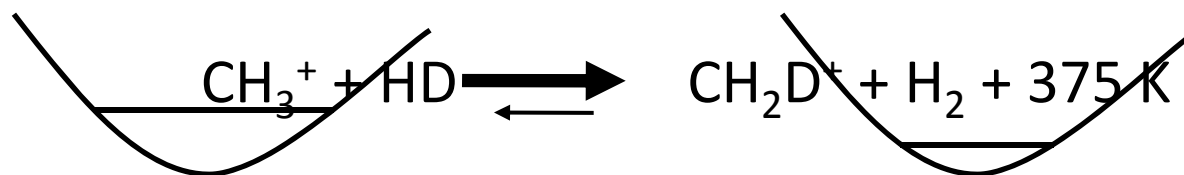
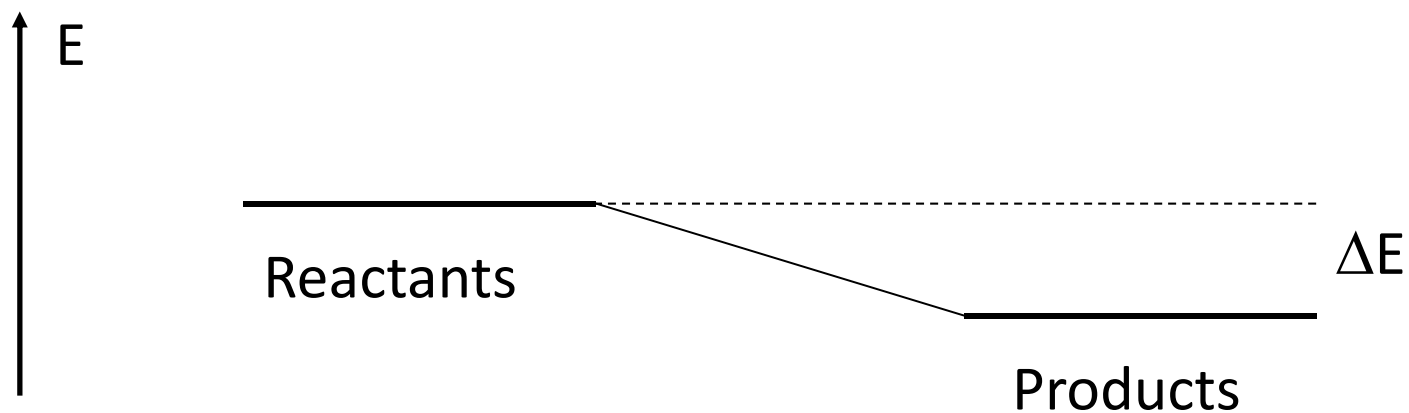
Isotope Enrichment

$[AD]/[AH] \sim 0.1$



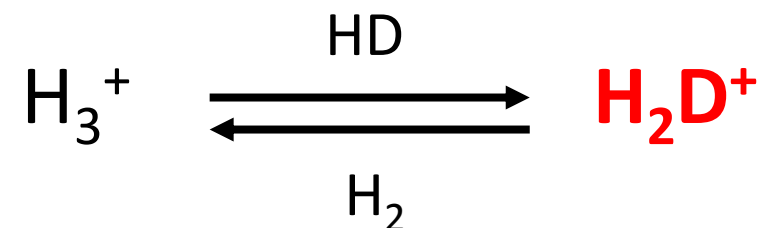
B. Parise, A. Belloche, F. Du, R. Güsten and K. Menten  
A&A, 526, A31 (2011)

# Primary Deuteration Reactions



# Isotopic Fractionation

## Ideal Case – Laboratory Situation

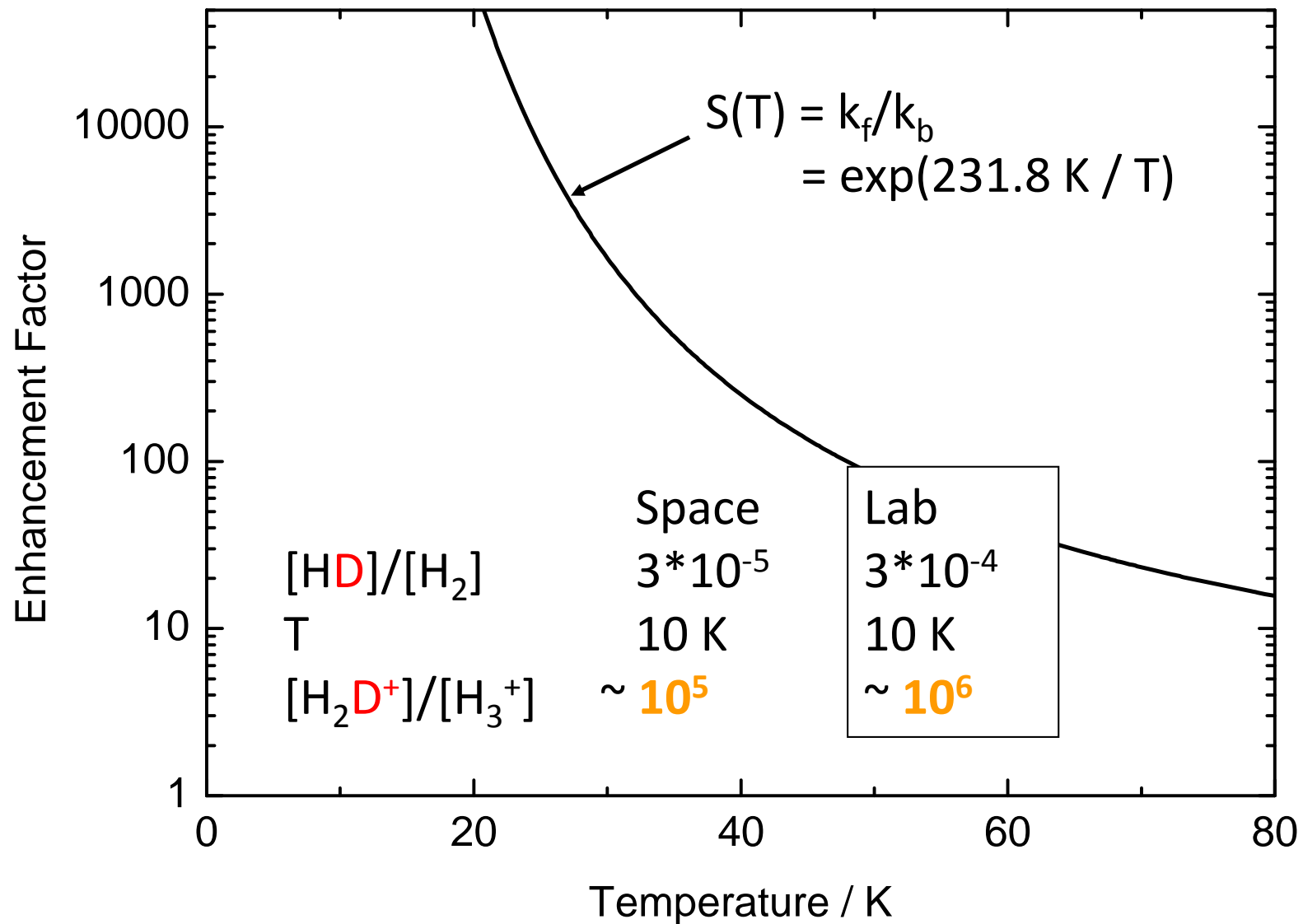


Equilibrium

$$[\text{H}_2\text{D}^+]/[\text{H}_3^+] = S(T) [\text{HD}]/[\text{H}_2]$$

$$S(T) = k_f/k_b$$

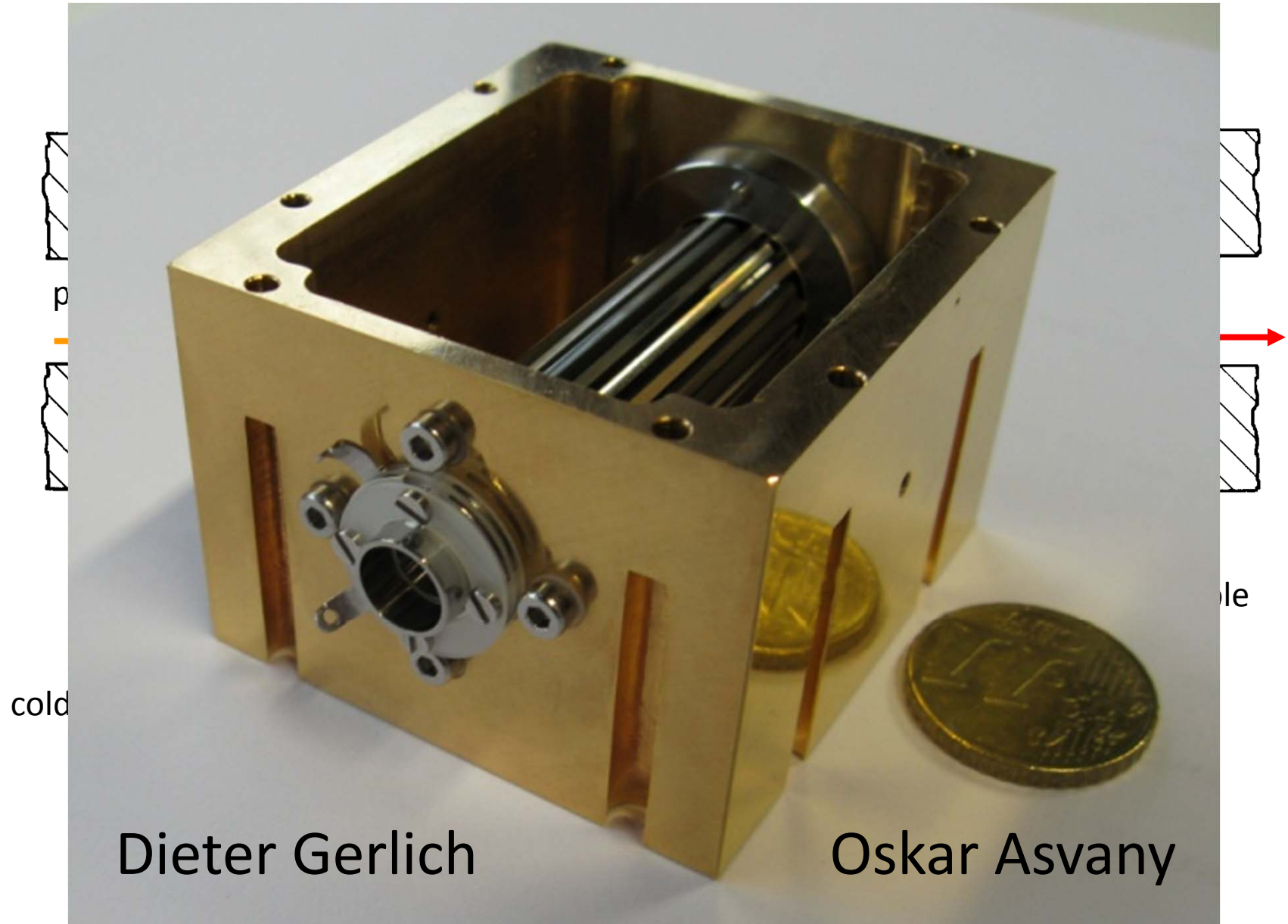
# Isotopic Fractionation





**Experimental Method:**  
**Electrodynamical Trapping**

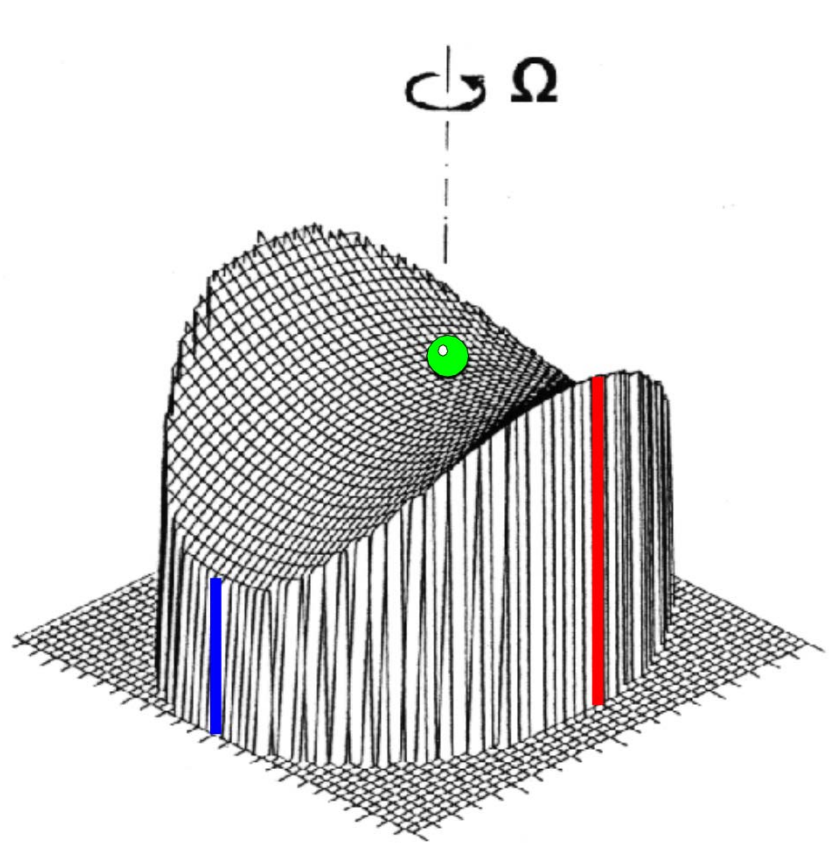
# 22-Pole Low Temperature Ion Trap



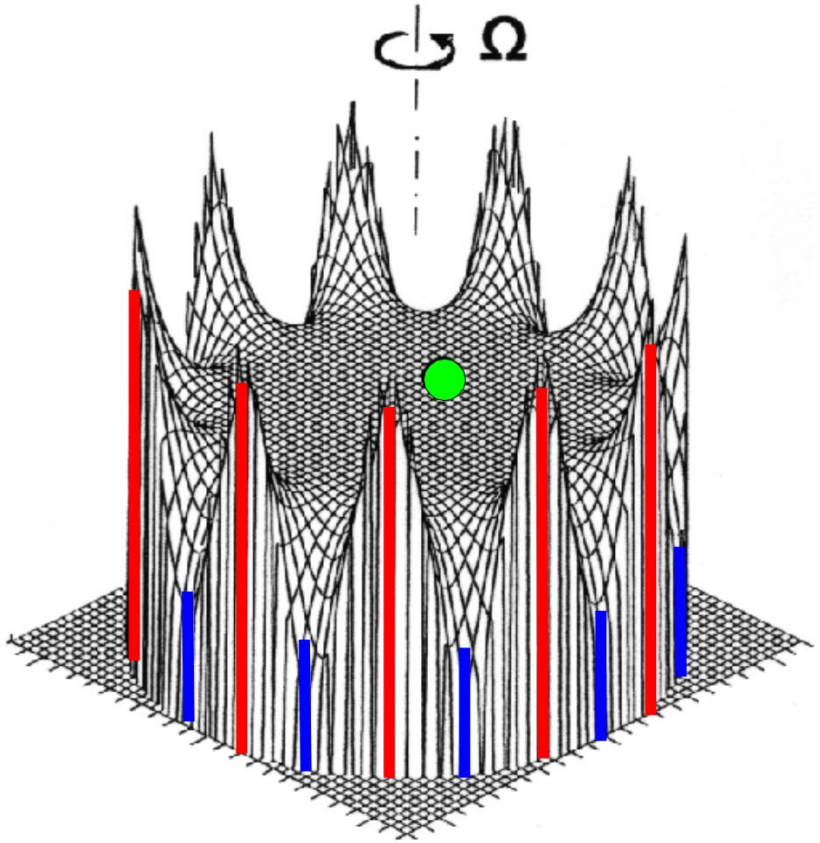
Dieter Gerlich

Oskar Asvany

# RF Ion Trap Mechanical Model

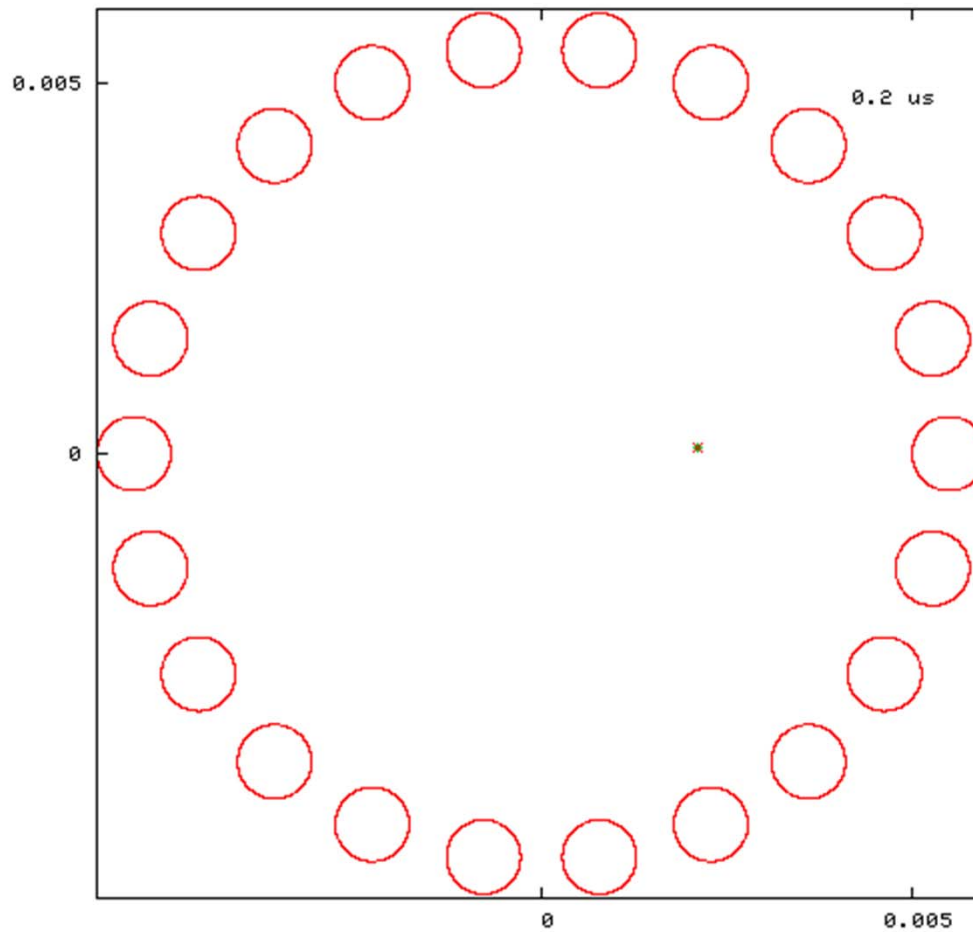


Quadrupole



22-Pole

# Trajectories of ions in 22-pole trap

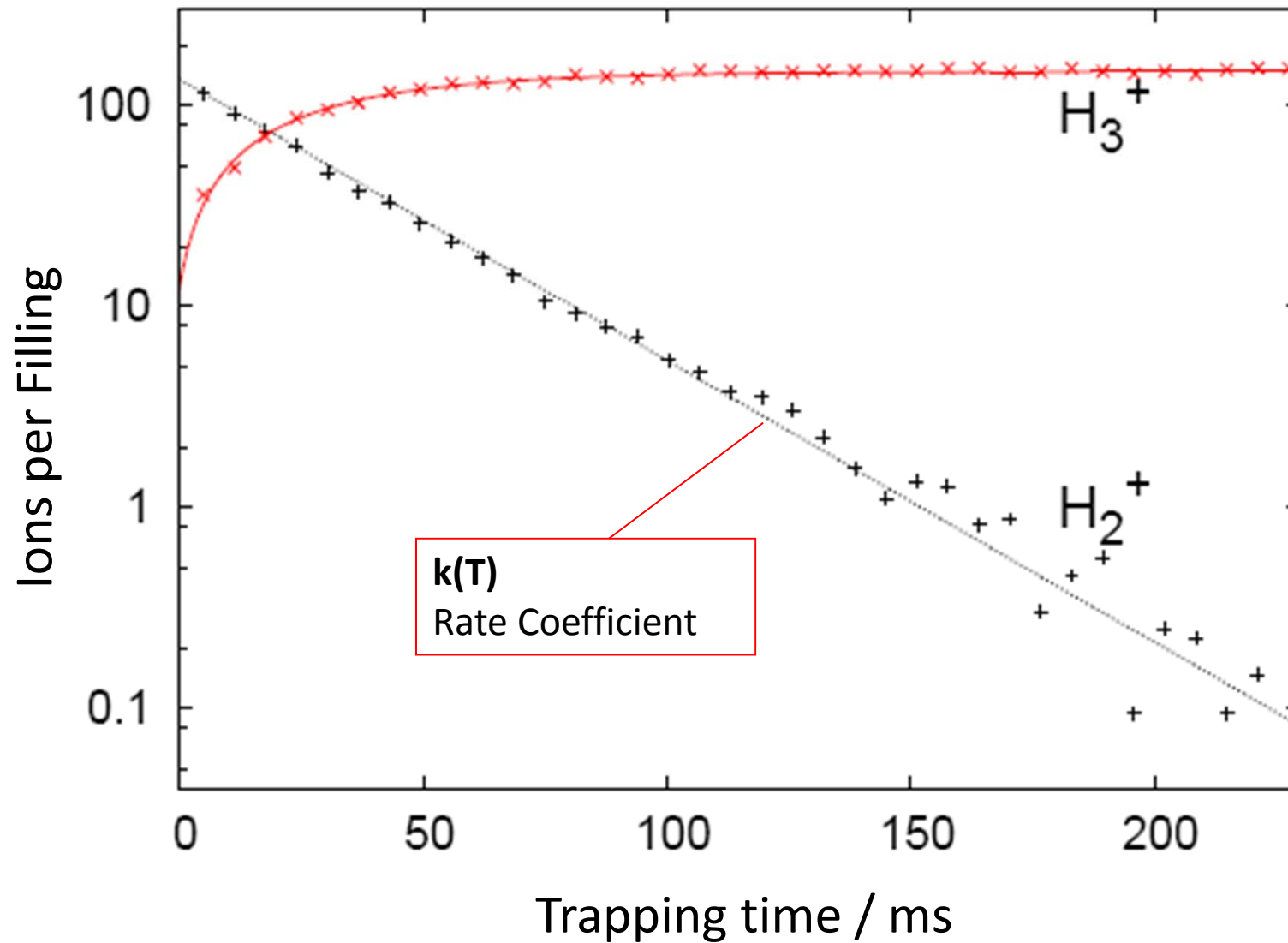
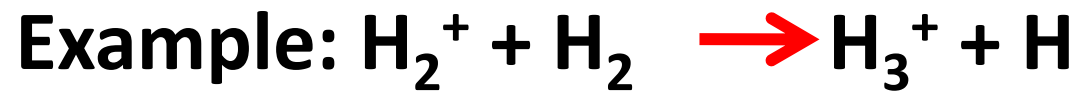


$f = 17\text{MHz}$

$V_0 = 20\text{V}$

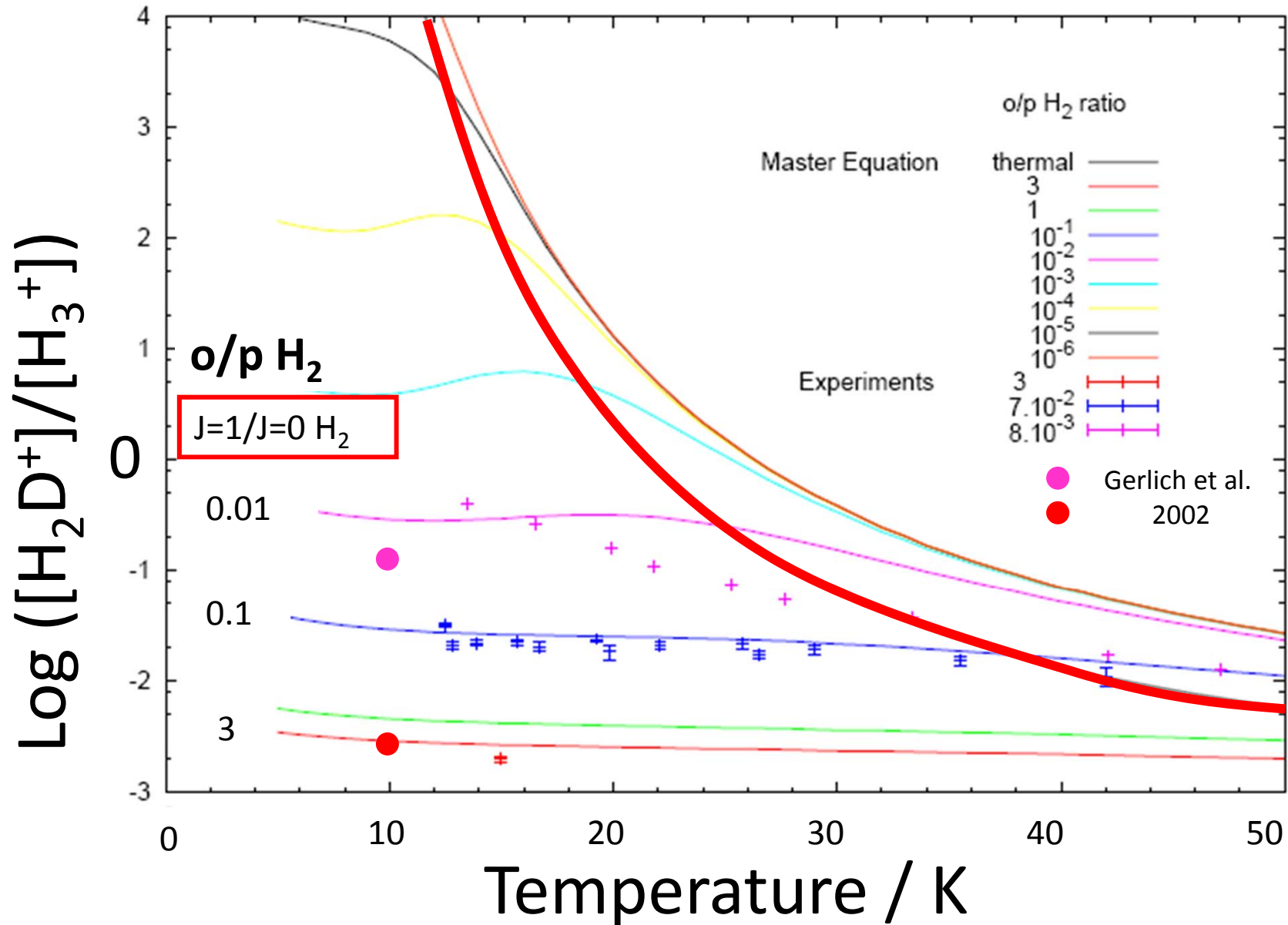
+  $m = 20u$

x  $m = 2u$

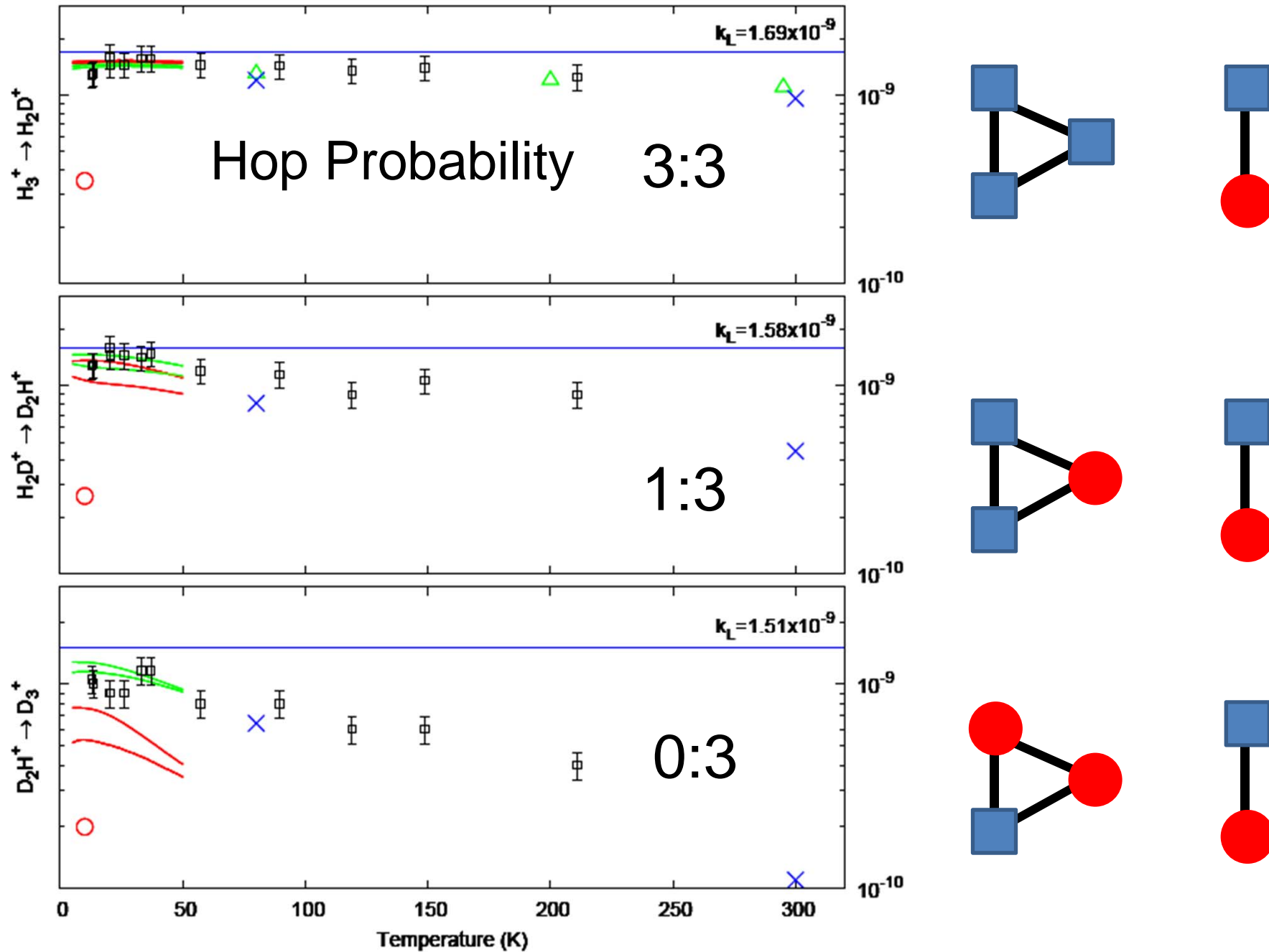


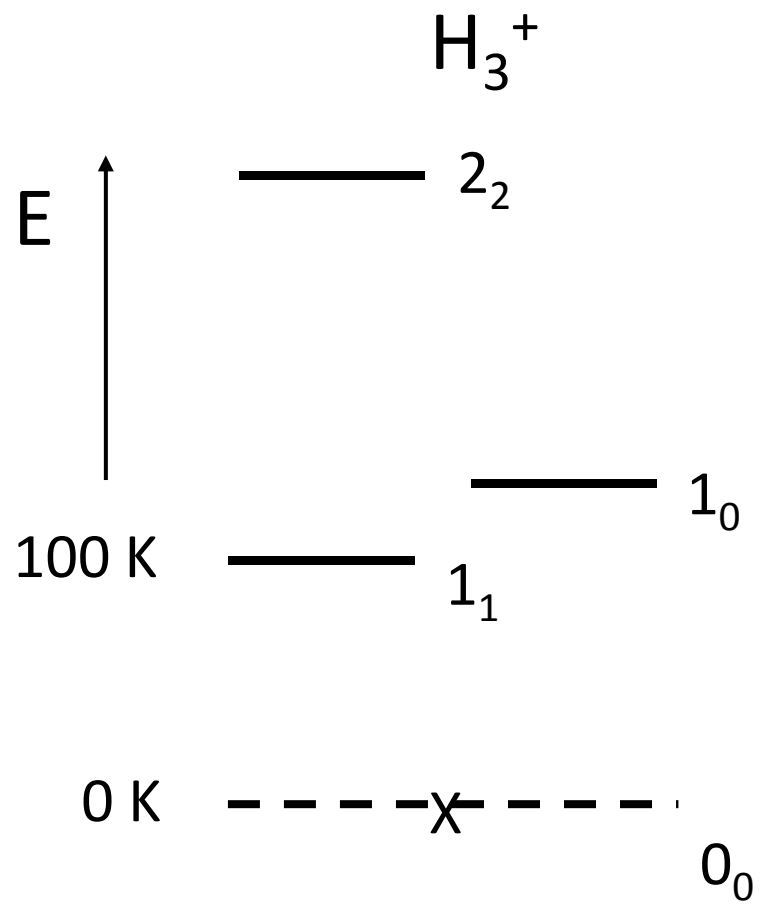
# New Experimental Results & Modelling

Hugo et al., J.Chem.Phys. 2009, **130**, 164302

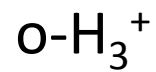
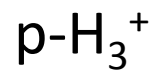


# Reaction Mechanism: Proton Hop

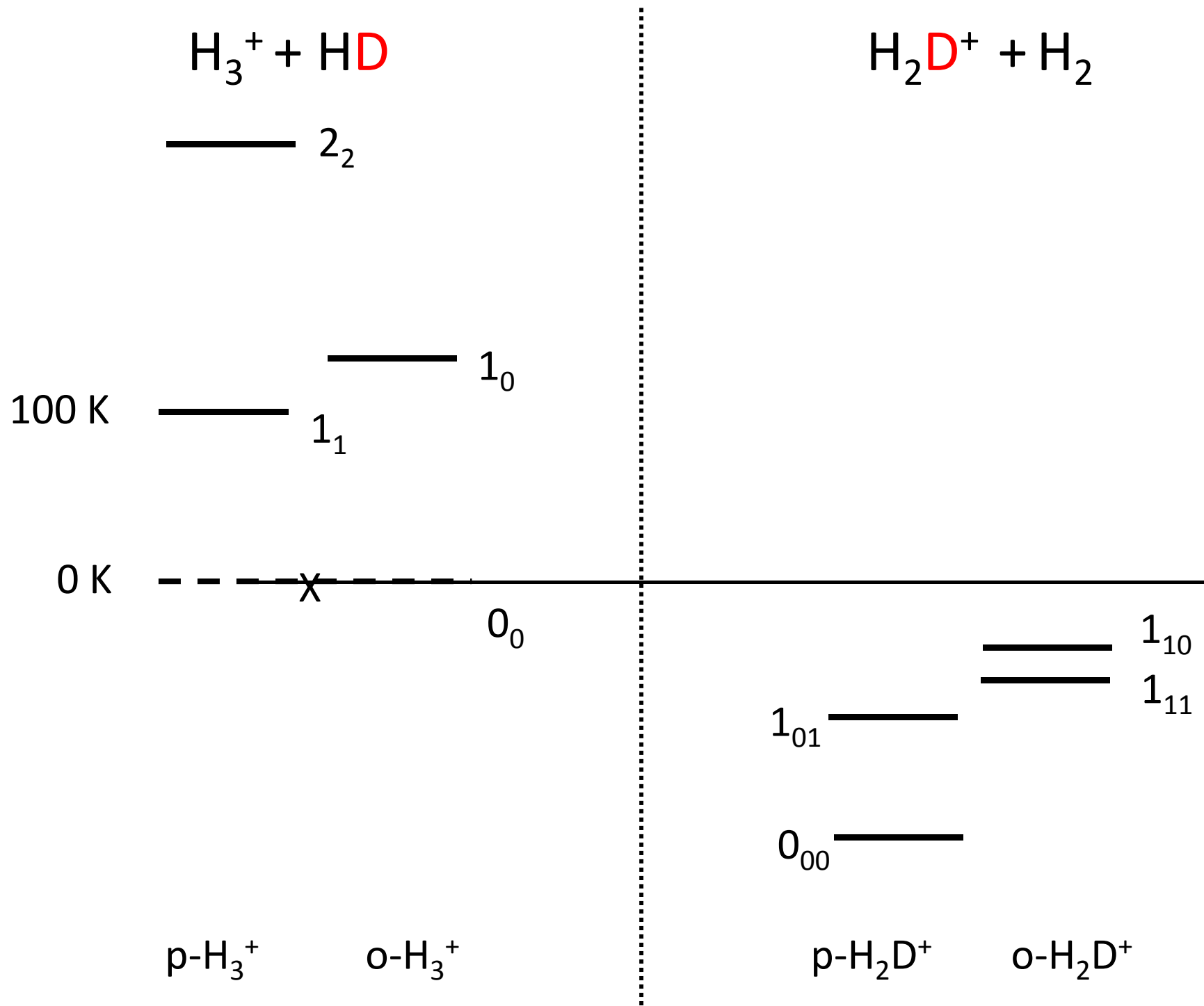


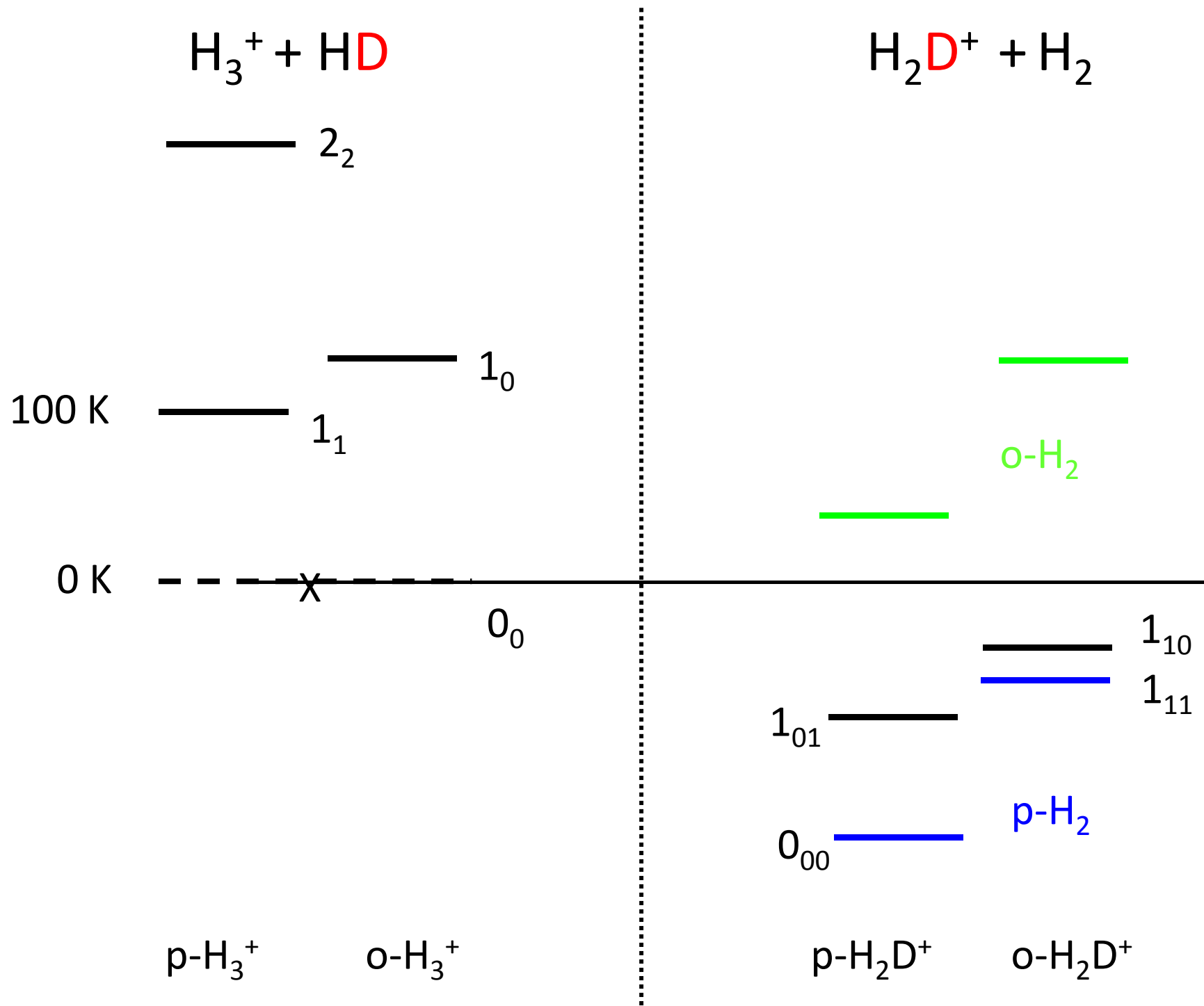


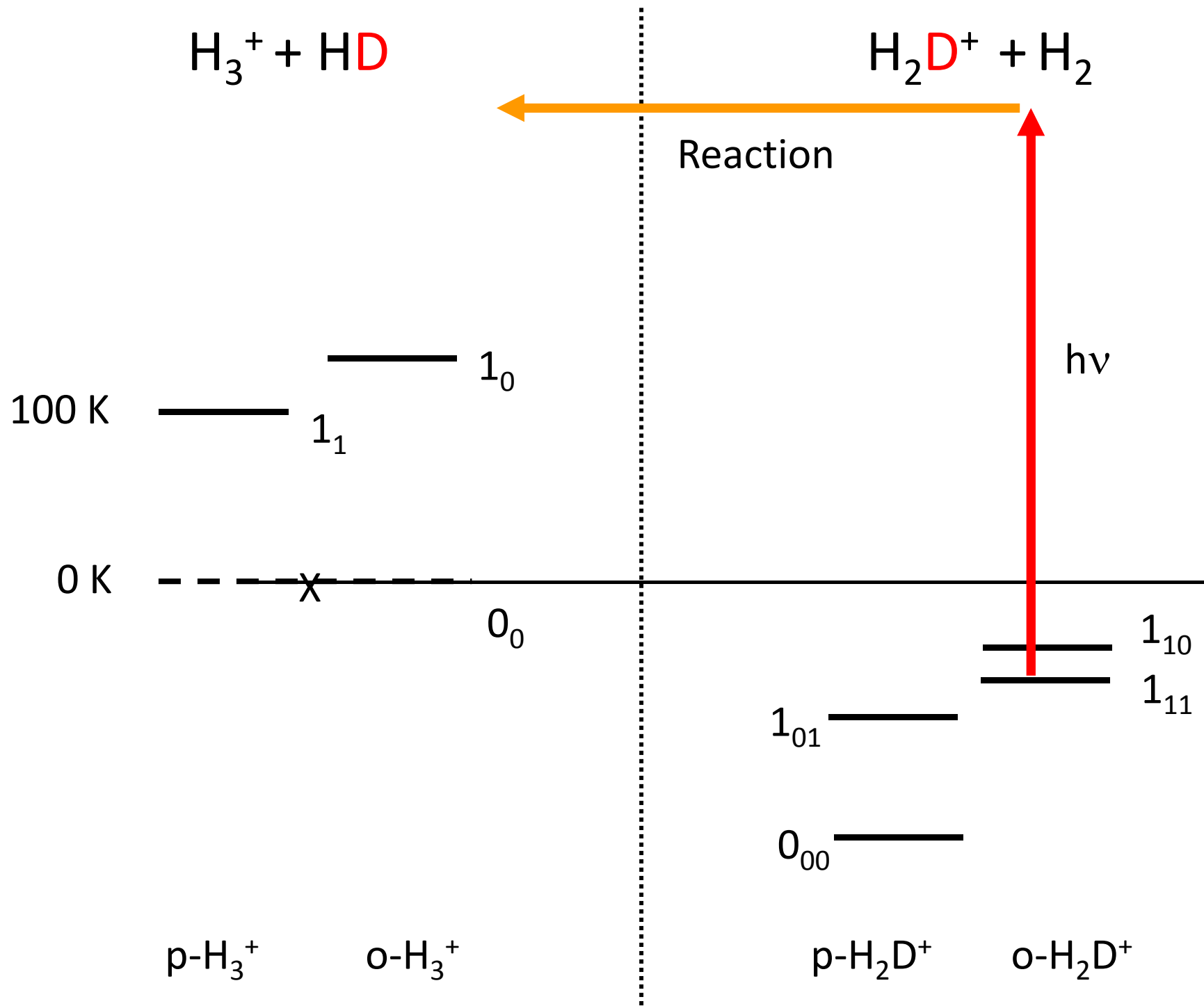
Lowest energy levels of  $H_3^+$

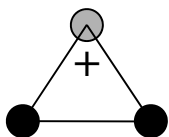




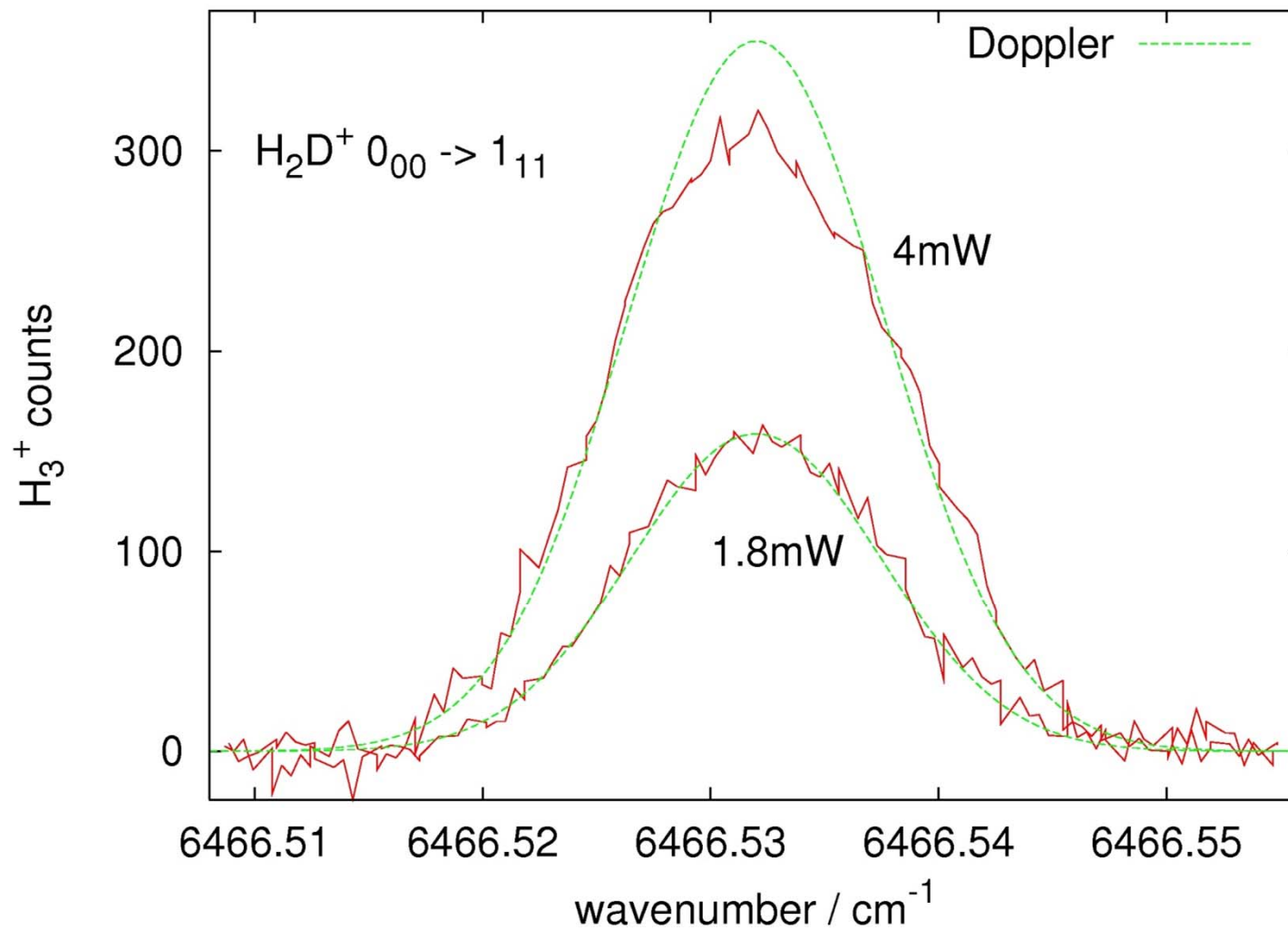








# Spectroscopic results for $\text{H}_2\text{D}^+$



$$T_{\text{Doppler}} = (27 \pm 2) \text{K}$$

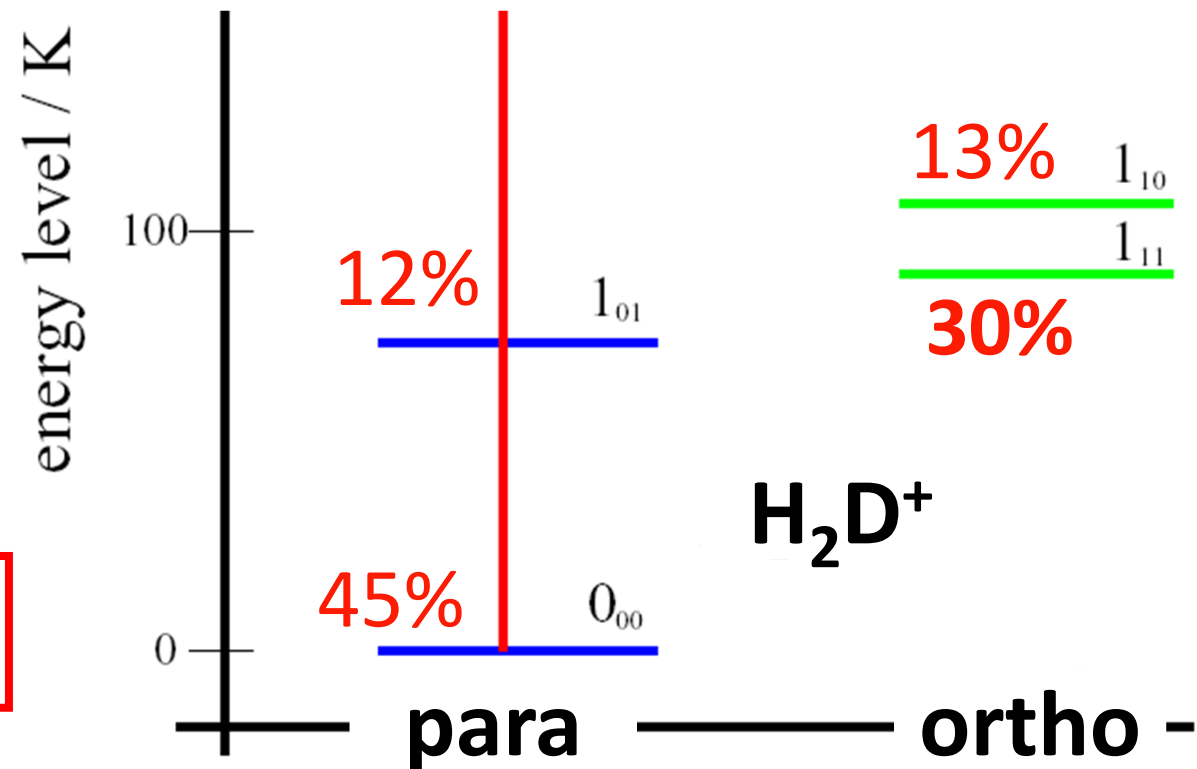
# rotational level populations of H<sub>2</sub>D<sup>+</sup>

$$\text{pop} \sim \frac{[H_3^+]}{B \cdot P}$$

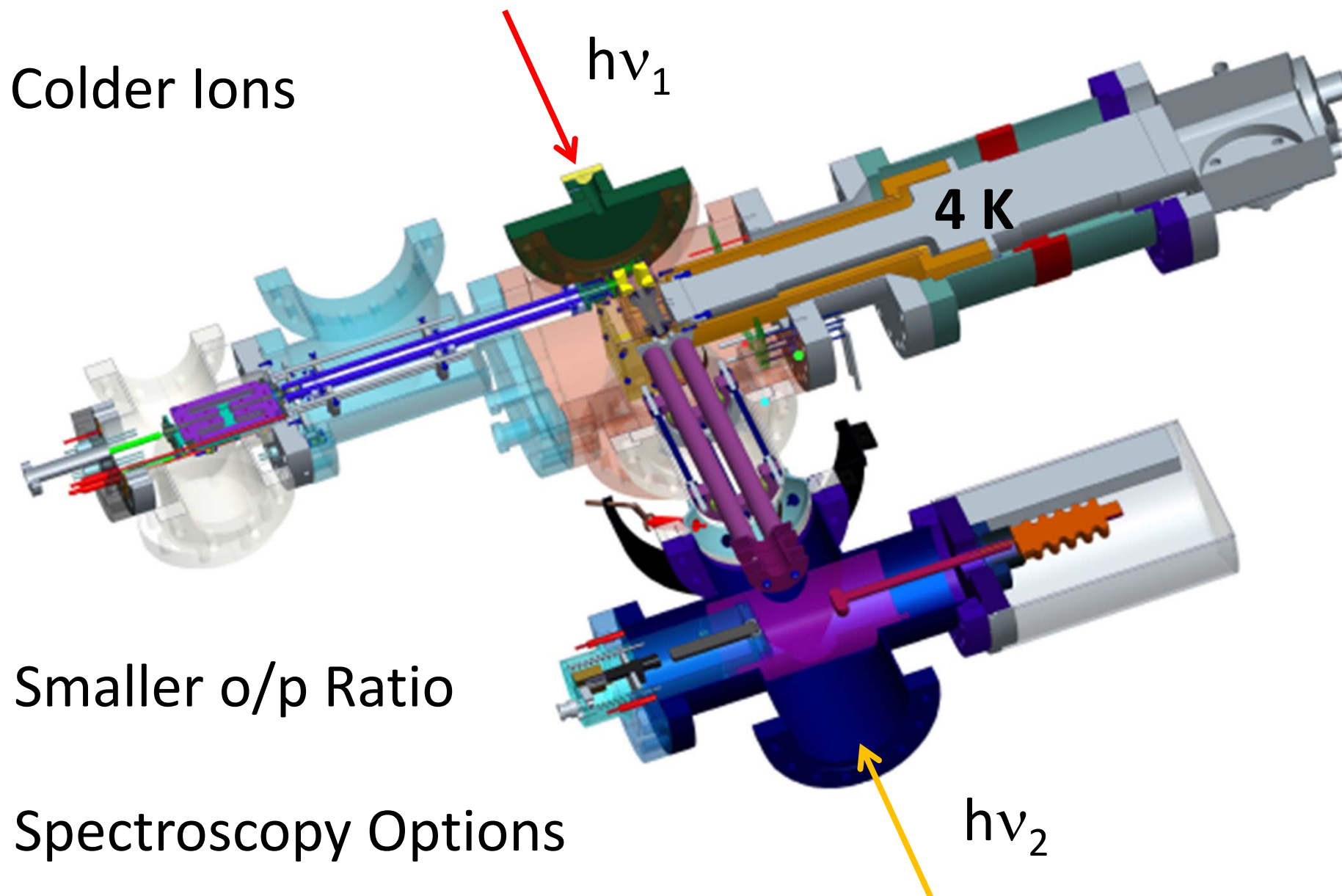
$$T_{\text{Doppler}} = (27 \pm 2) \text{K}$$

$$T_{\text{rot,para}} = (27 \pm 2) \text{K}$$

$$T_{\text{ortho/para}} = (35) \text{K}$$

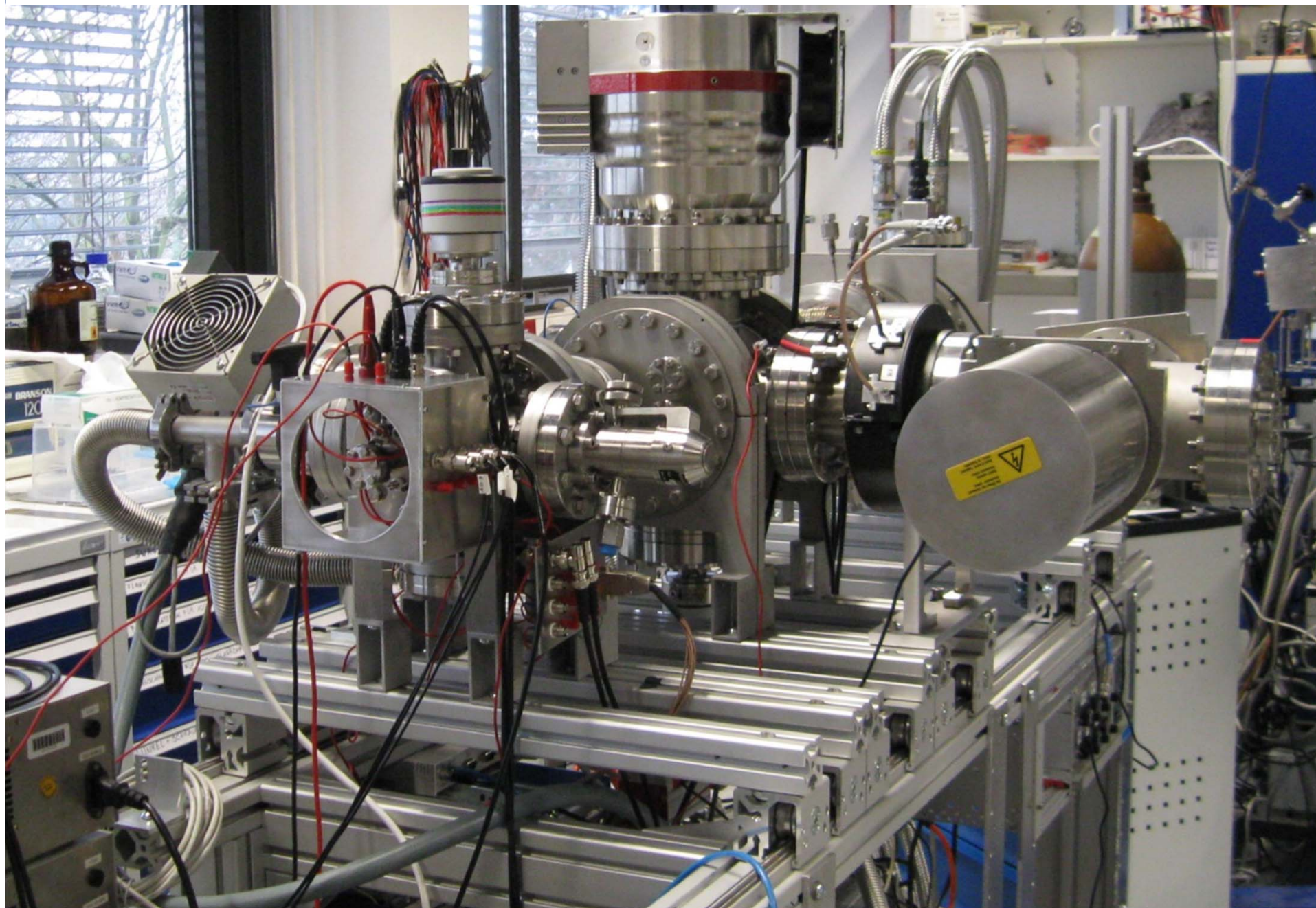


# COLTRAP: Future $\text{H}_3^+$ / $\text{H}_2\text{D}^+$ Experiments





# FELION: Spectroscopy Trap for FELIX



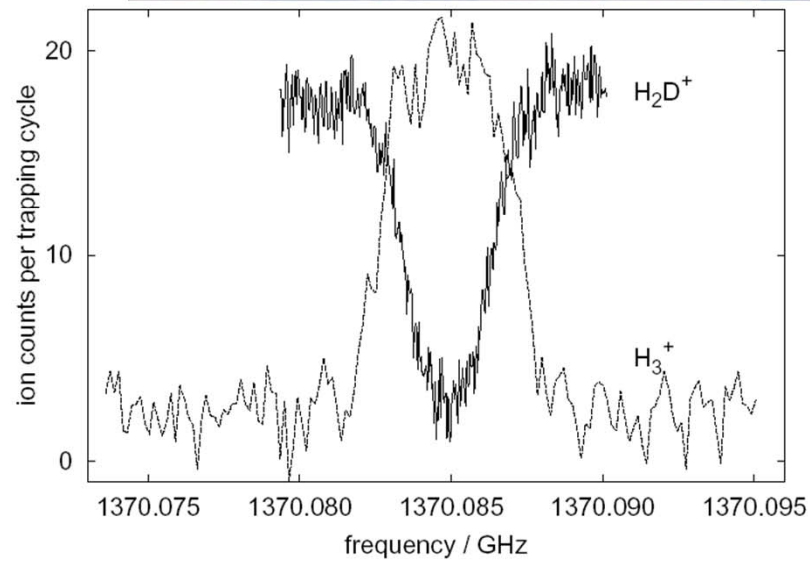


Li

# SOFIA

100 K

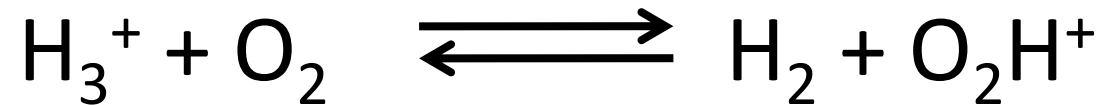
0 K





# IS HO<sub>2</sub><sup>+</sup> A DETECTABLE INTERSTELLAR MOLECULE?

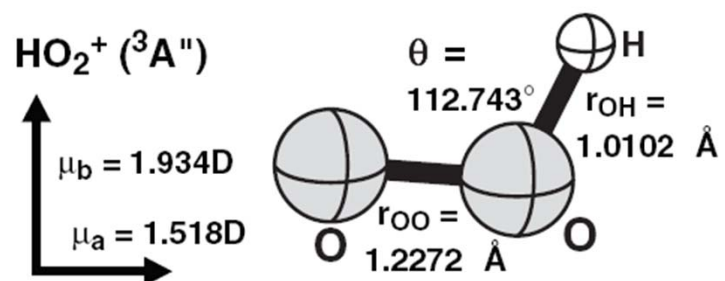
SUSANNA L. WIDICUS WEAVER<sup>1,4</sup>, DAVID E. WOON<sup>2</sup>, BRANKO RUSCIC<sup>3</sup>, AND BENJAMIN J. MCCALL



Tracer for O<sub>2</sub>?

Chemistry of near thermoneutral Reaction

# IS HO<sub>2</sub><sup>+</sup> A DETECTABLE INTERSTELLAR MOLECULE?

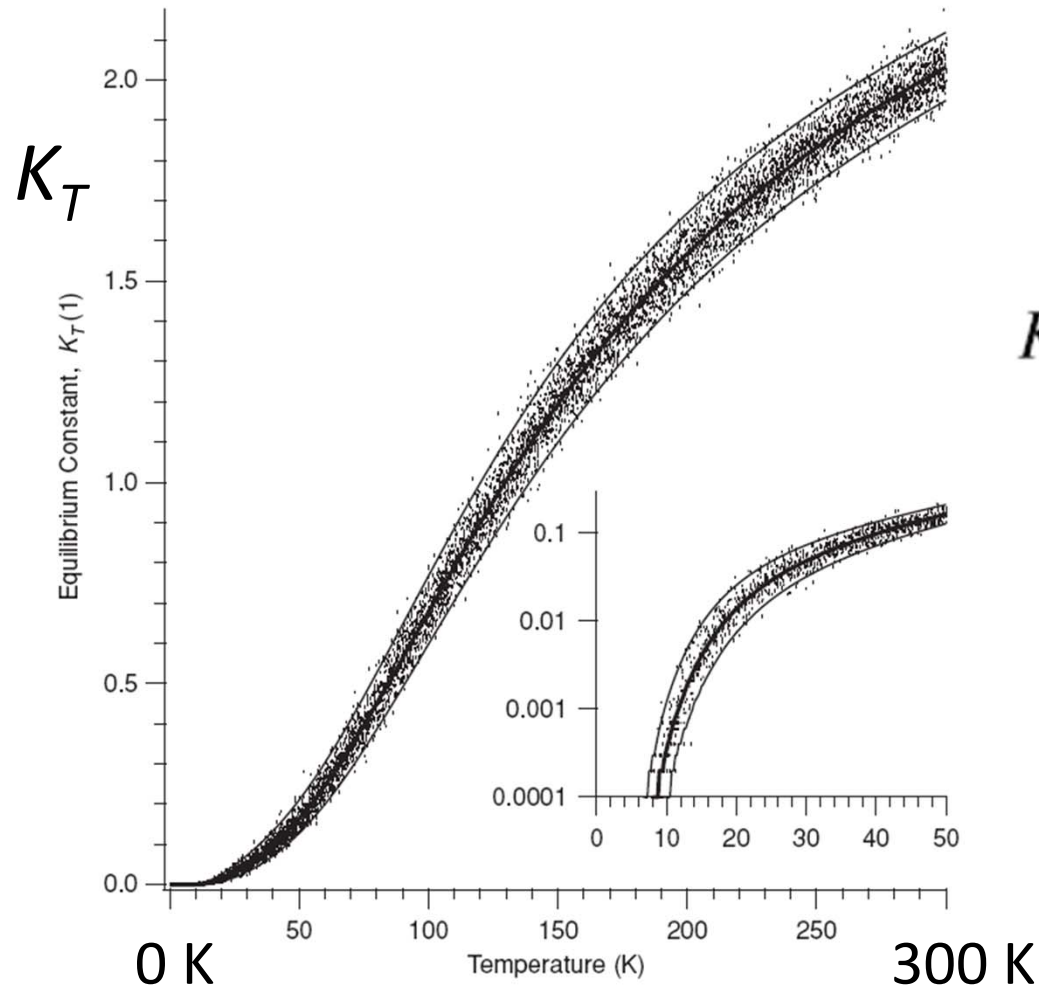
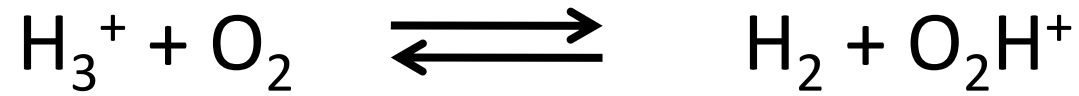


**Figure 1.** Equilibrium structural parameters for HO<sub>2</sub><sup>+</sup> (RCCSD(T) at the valence complete basis set limit with core–valence corrections) and dipole moment components (MRCI/AV5Z).

Molecular Parameters Determined for HO<sub>2</sub> and HO<sub>2</sub><sup>+</sup>

Spectroscopic Constant	HO <sub>2</sub> Calculated <sup>a</sup>	HO <sub>2</sub> Experimental <sup>b</sup>	HO <sub>2</sub> <sup>+</sup> Calculated <sup>a</sup>
$A_0$ (GHz)	615.997	610.2733	659.301
$B_0$ (GHz)	33.604	33.5178	38.344
$C_0$ (GHz)	31.643	31.6677	35.885
$\nu_1$ (cm <sup>-1</sup> )	3457	3436.2	3028
$\nu_2$ (cm <sup>-1</sup> )	1406	1391.8	1440
$\nu_3$ (cm <sup>-1</sup> )	1128	1097.6	1068
$\Delta_N$ (MHz)	0.1127	0.116908	0.1075
$\Delta_{NK}$ (MHz)	3.303	3.44572	5.515
$\Delta_K$ (MHz)	115.02	123.5906	299.03
$\mu_a$ (D)	1.405	1.412	1.518
$\mu_b$ (D)	1.572	1.541	1.934
$D$ (cm <sup>-1</sup> )			6.870
$E$ (cm <sup>-1</sup> )			0.033
$\epsilon_{aa}$ (MHz)	-46730	-49572	-1182
$\epsilon_{bb}$ (MHz)	-432	-422.9	-481
$\epsilon_{cc}$ (MHz)	-159	8.748	-476

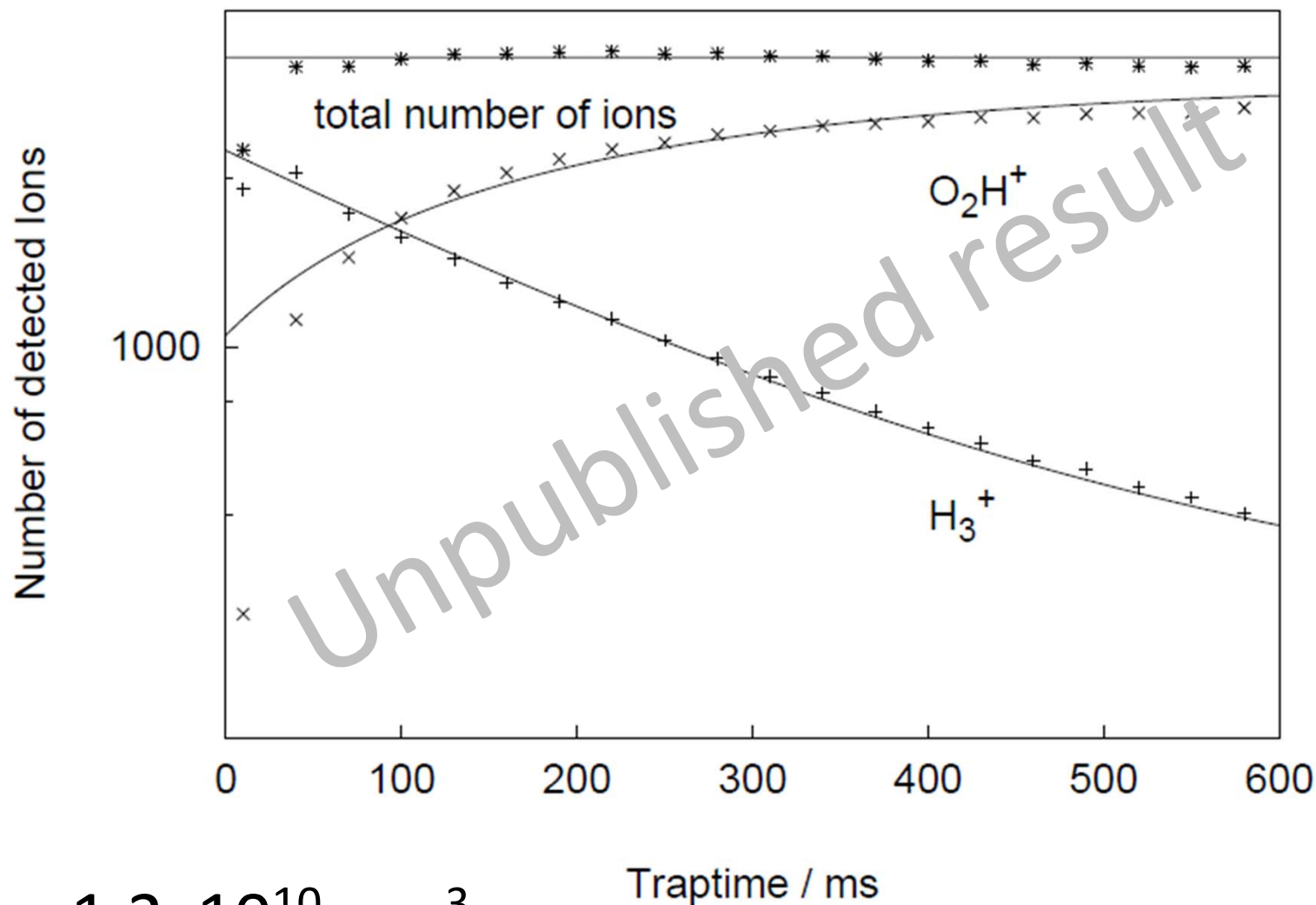
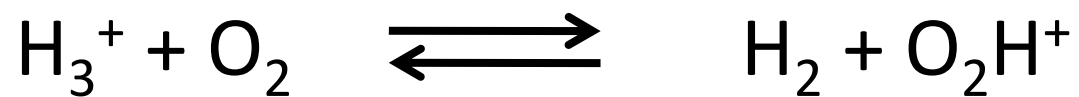
# Predictions: Equilibrium Constant



$$K_T(1) = e^{-\frac{E_0(1)}{kT}} \frac{q(\text{HO}_2^+)q(\text{H}_2)}{q(\text{H}_3^+)q(\text{O}_2)}$$

Formation of  $\text{O}_2\text{H}^+$  endothermic:  $E_a/k = 72 \pm 13 \text{ K}$

# Trap Experiments



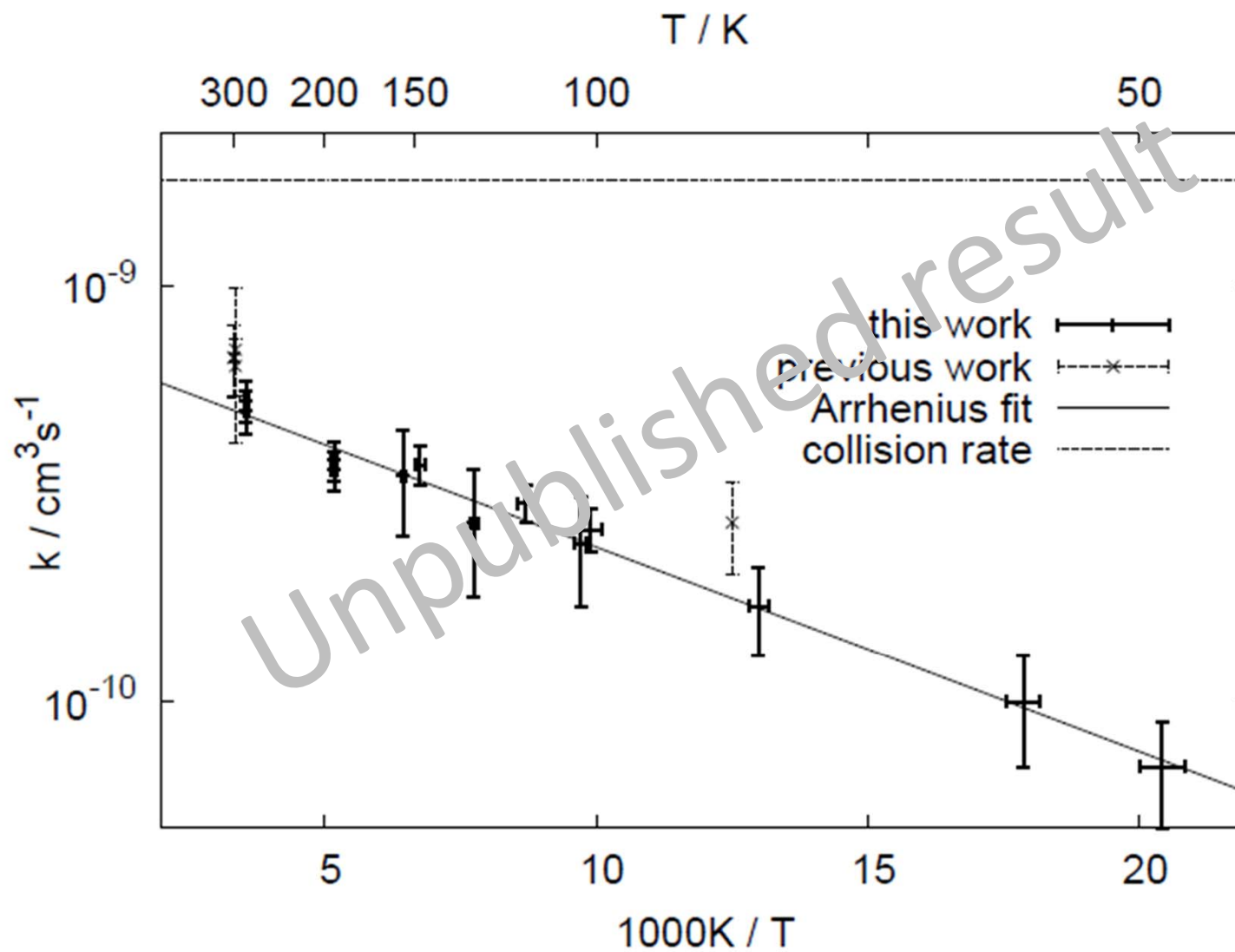
$[\text{O}_2] = 1.2 \times 10^{10} \text{ cm}^{-3}$

Trap time / ms

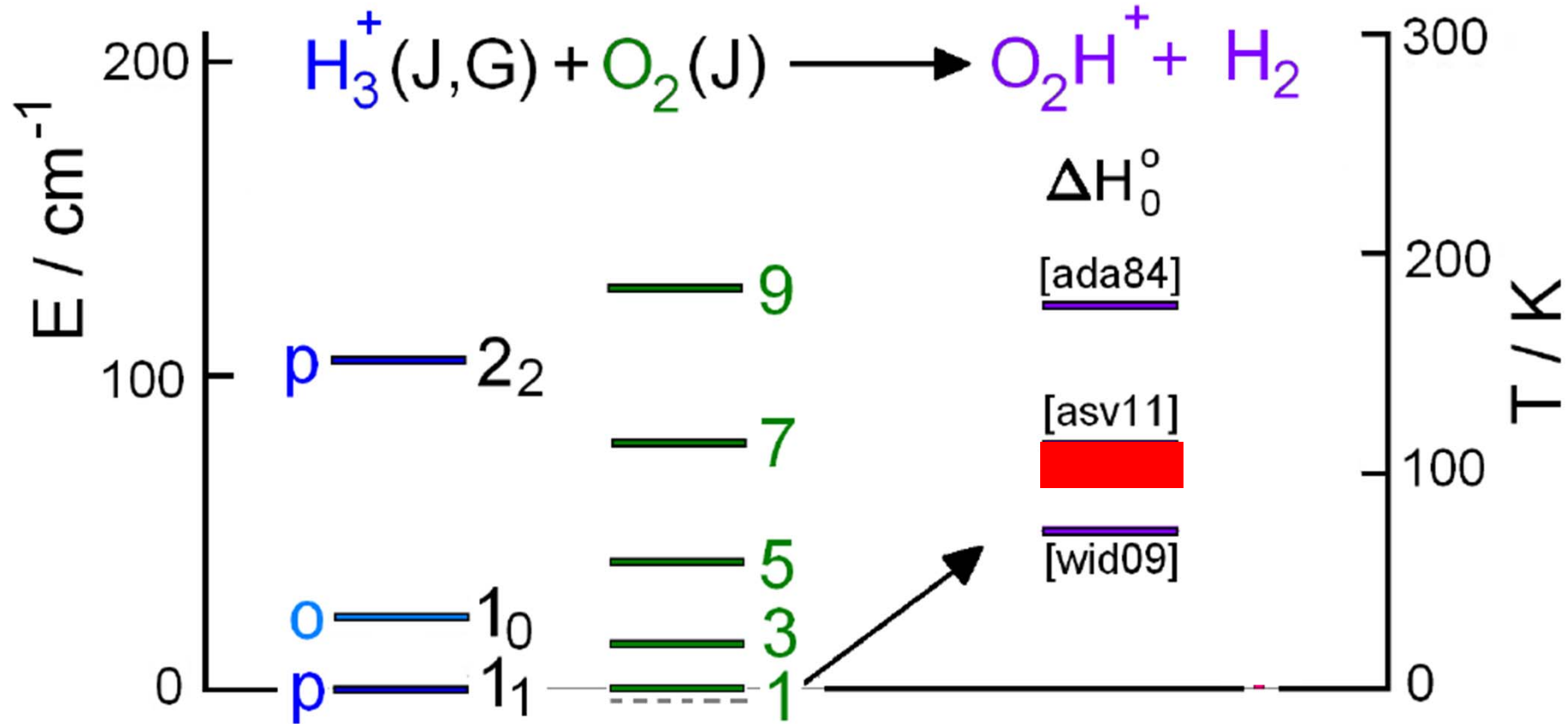
T = 129 K

# Arrhenius Plot

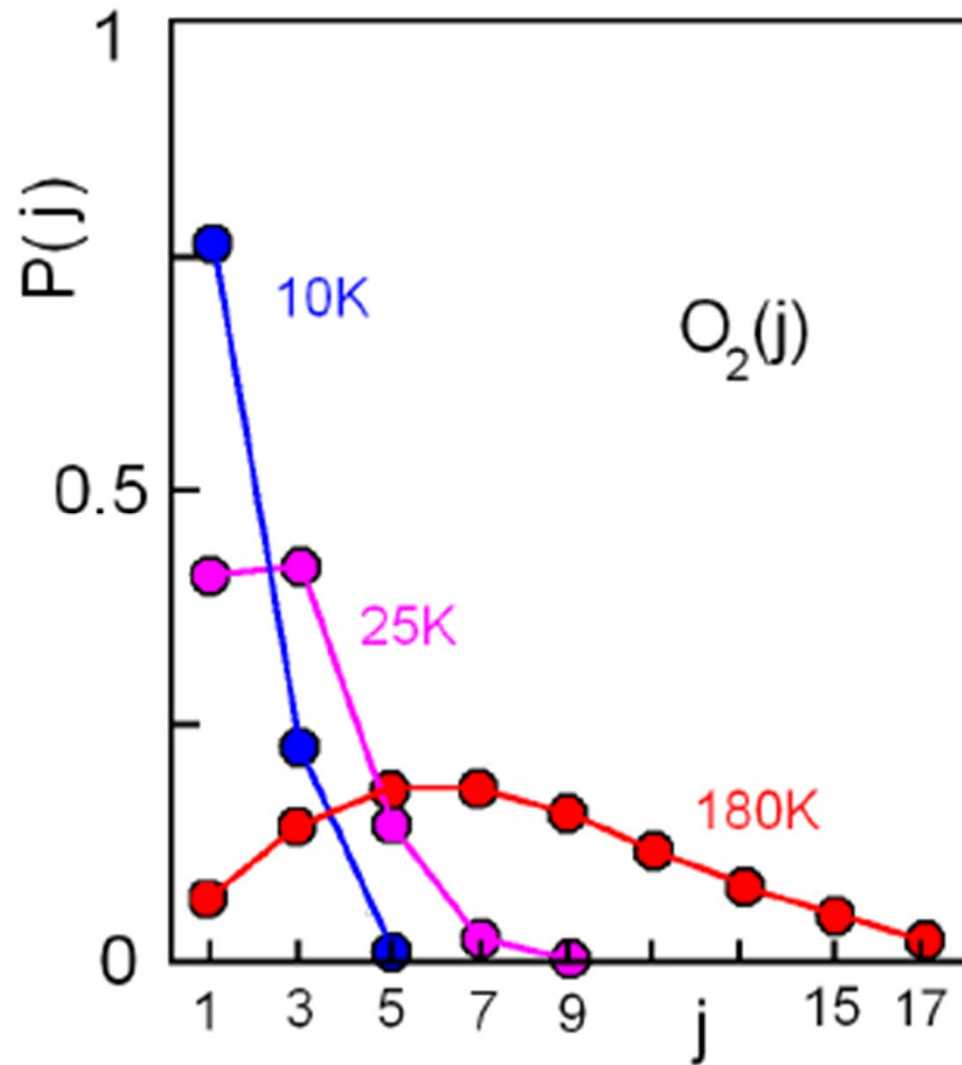
$$E_a/k = 113 \pm 4 \text{ K}$$



# Energetics of $\text{H}_3^+ + \text{O}_2$ Collision

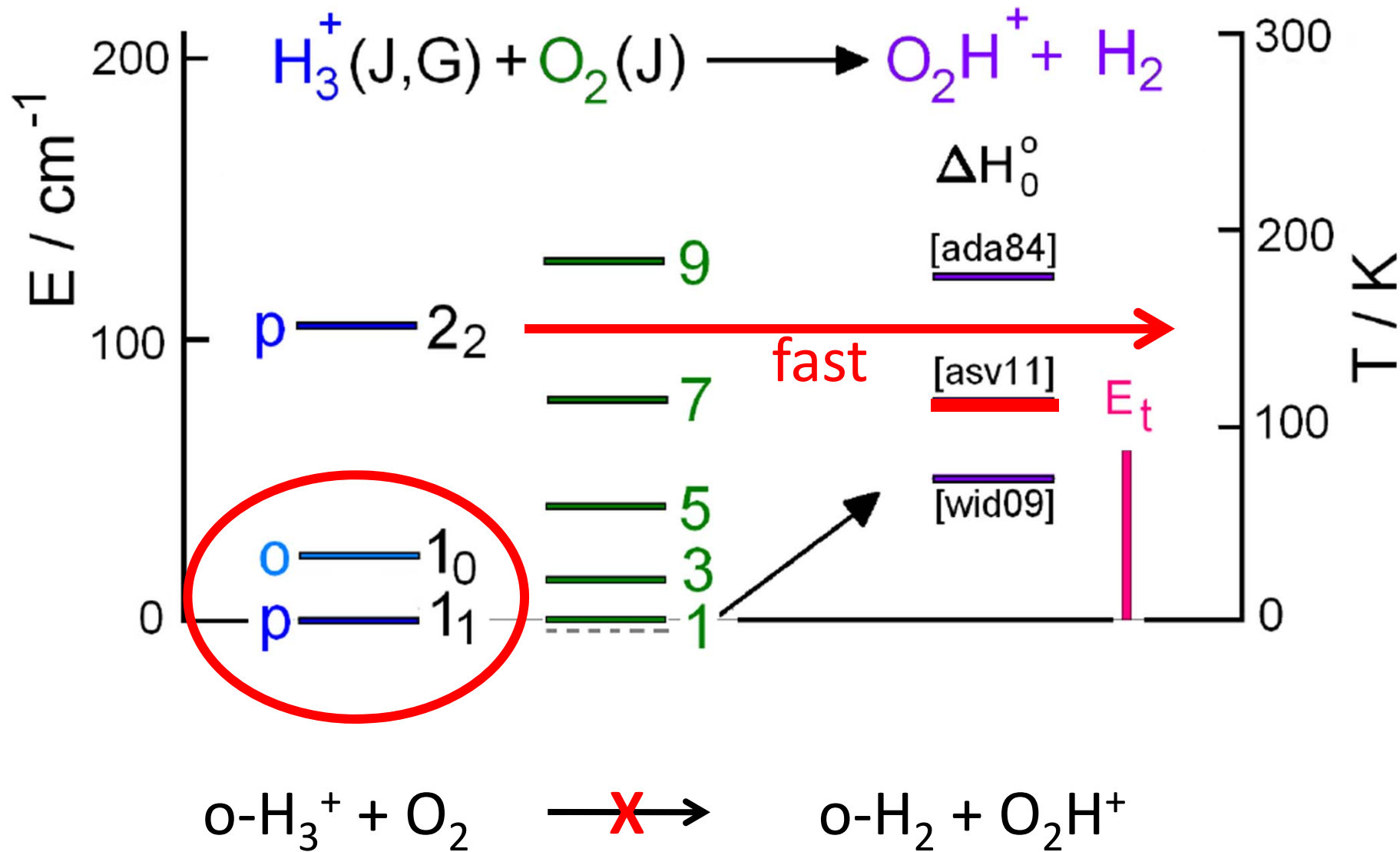


# O<sub>2</sub> Rotational State Distribution



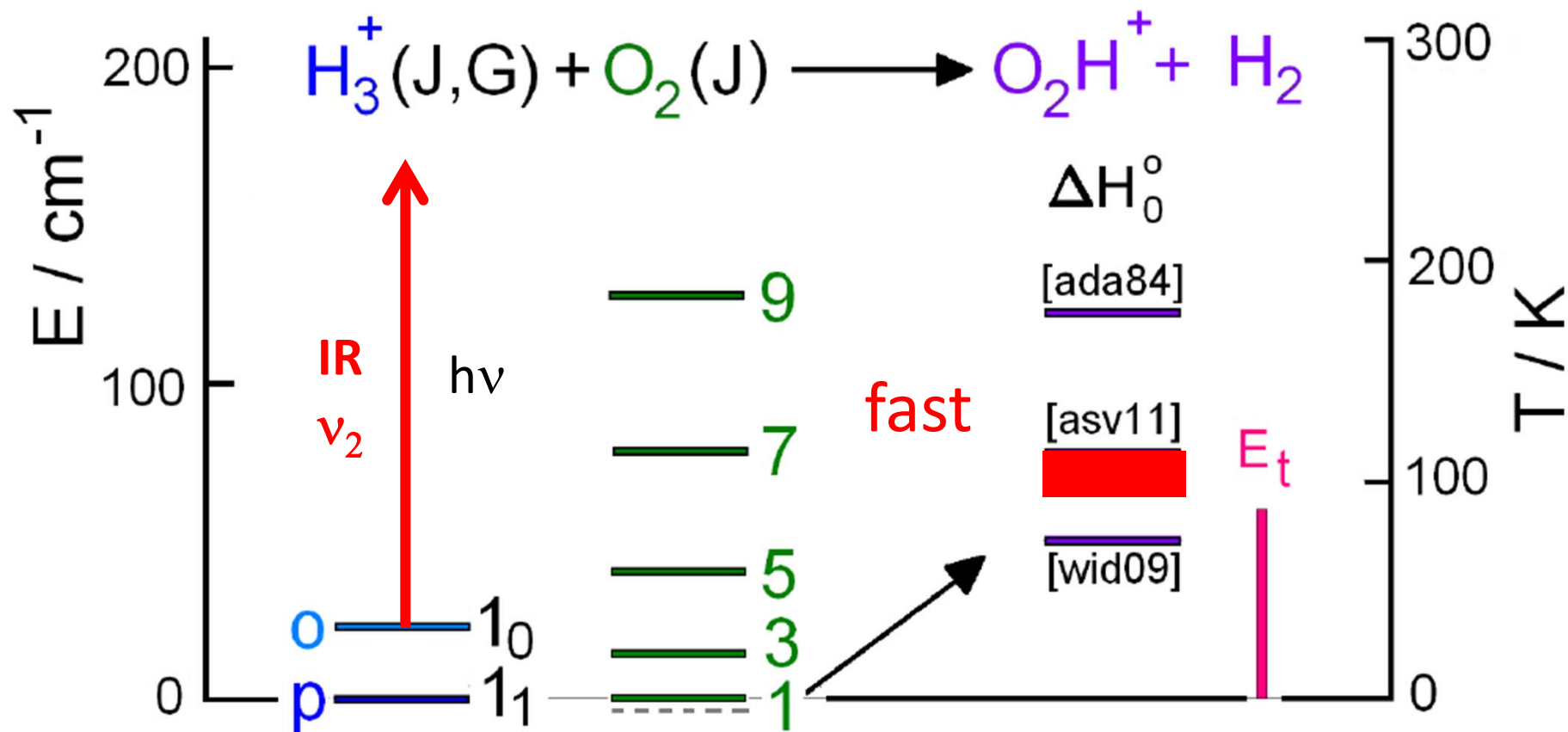
Dieter Gerlich at 2011 COST Meeting

# Towards State Preparation: Collaboration with Dieter Gerlich





# State Population: Light Induced Reactions



# Observation of the Infrared Spectrum of $\text{H}_3^+$

Takeshi Oka

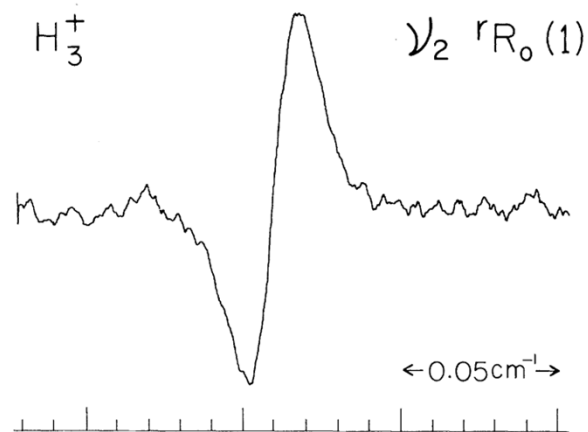
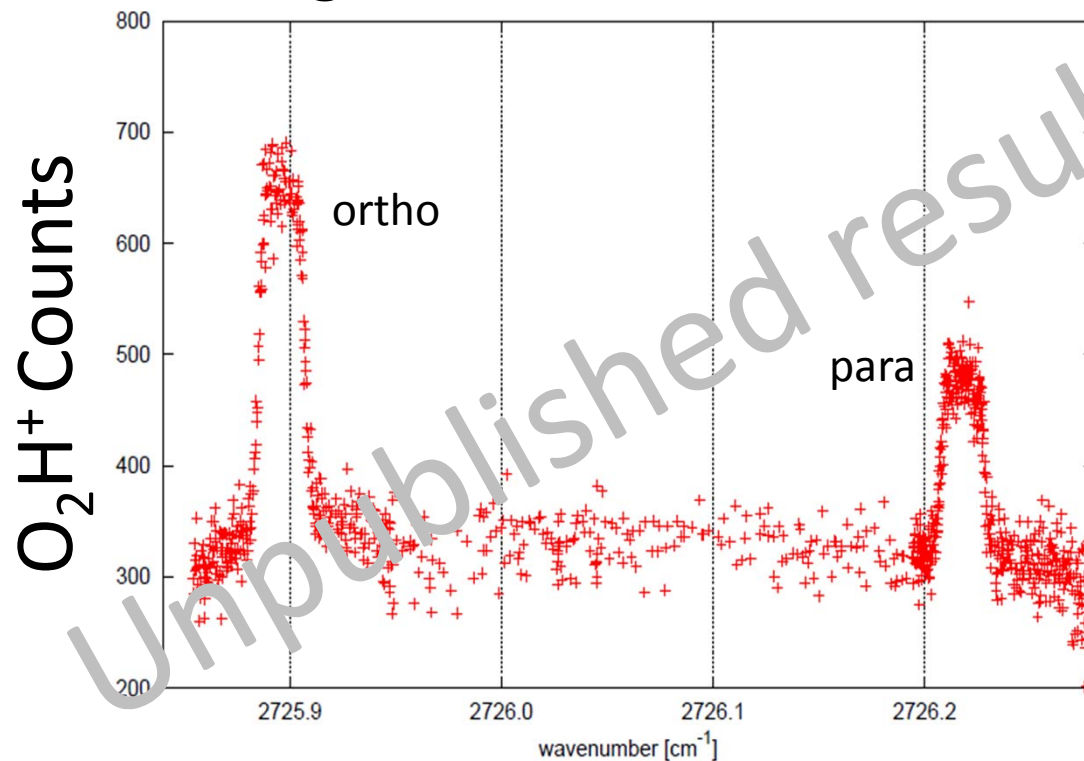
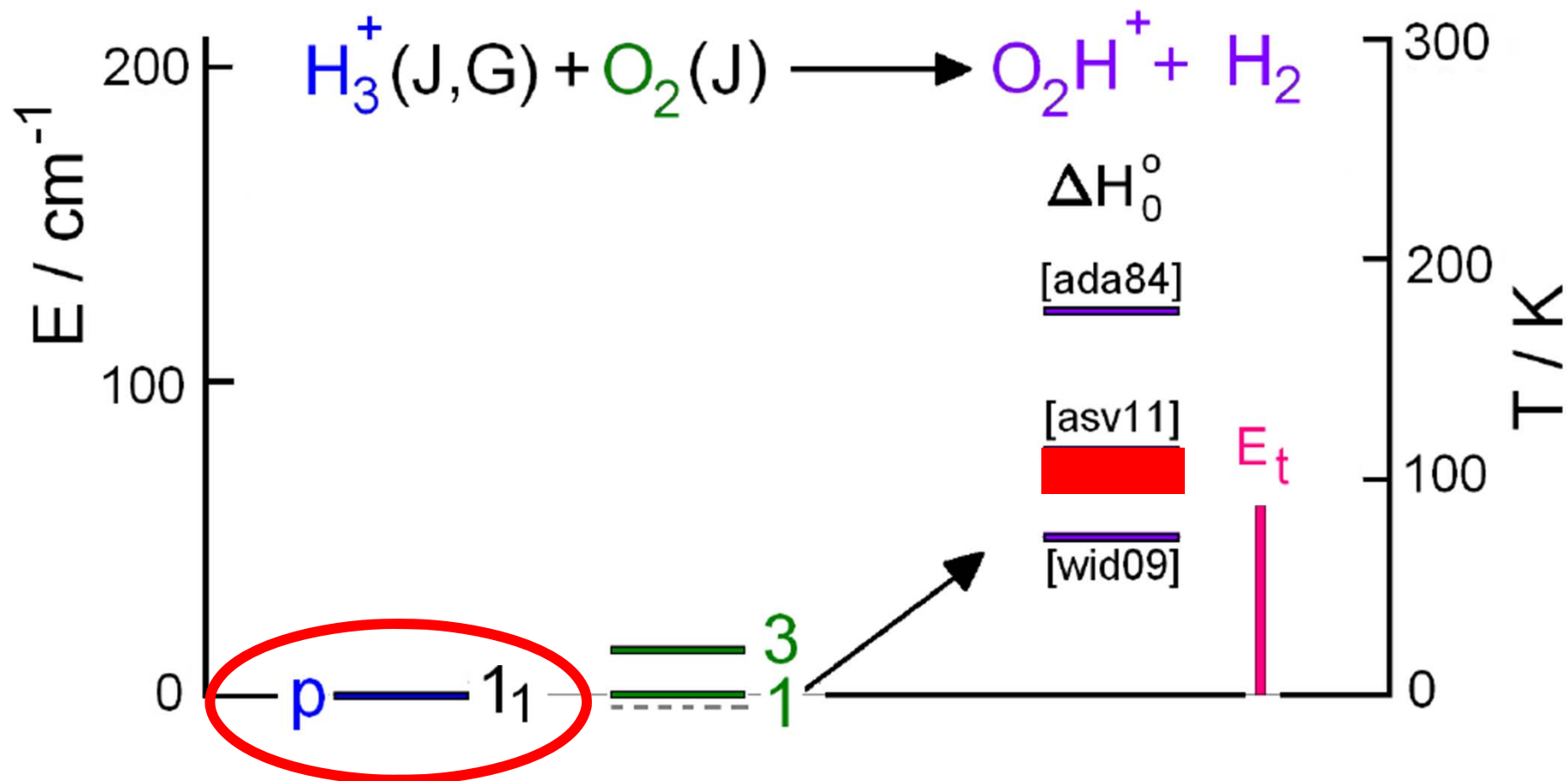


FIG. 1. The observed  $rR_0(1)$  transition ( $J=2$ ,  $k=\pm 1$ ,  $l=\pm 1$ )  $\leftarrow$  ( $J=1$ ,  $k=0$ ) of  $\text{H}_3^+$  at  $2725.885\text{ cm}^{-1}$ . Discharge current, 140 mA; pressure, 0.5 Torr; the time constant for detection, 400 msec.

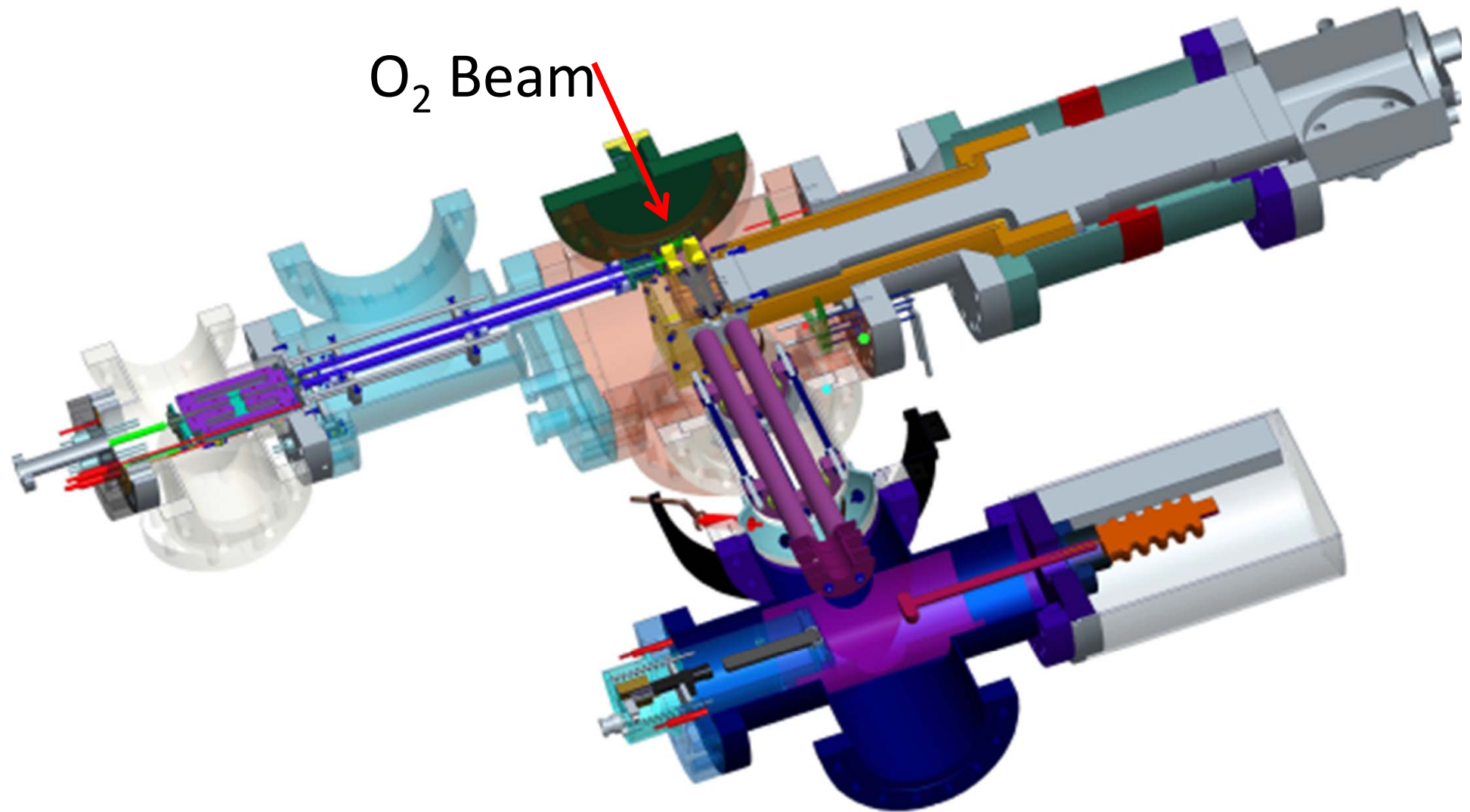
## Light Induced Reactions



# Towards State Preparation: Collaboration with Dieter Gerlich



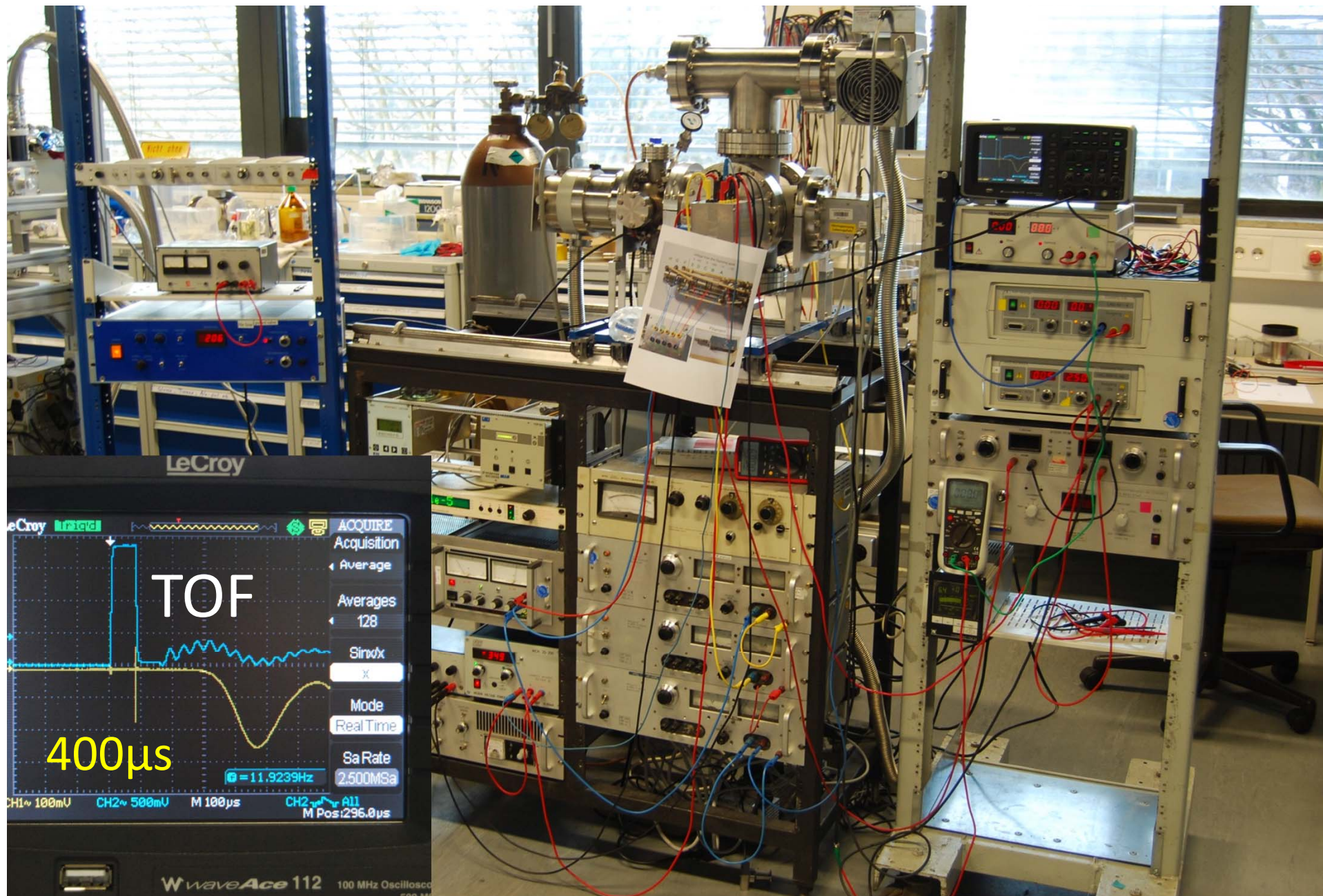
# COLTRAP: Combined Molecular Beam & Trap



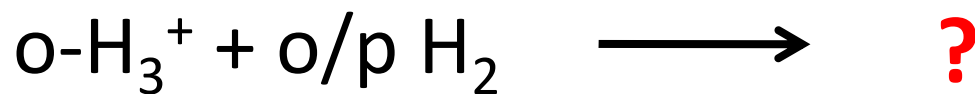
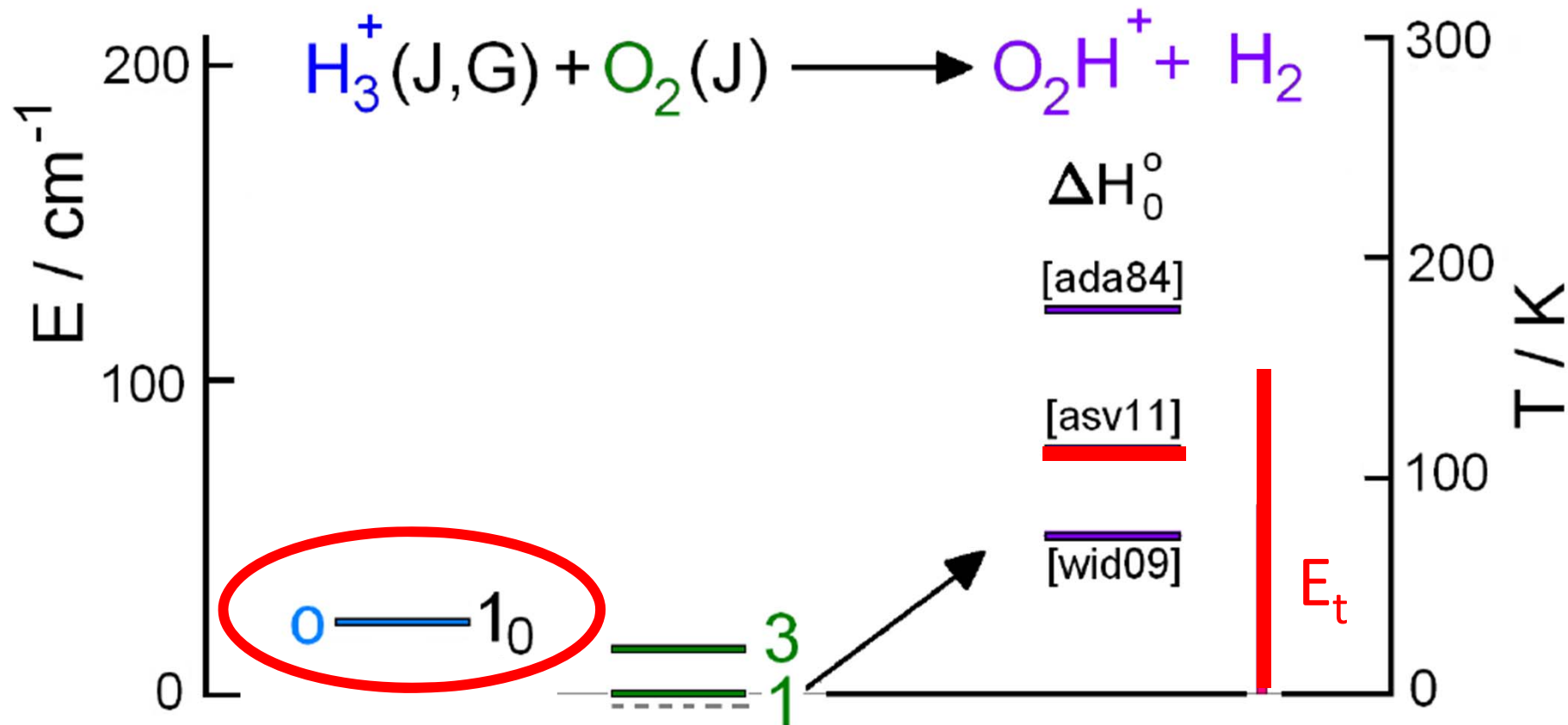
Towards State Preparation: Collaboration with Dieter Gerlich



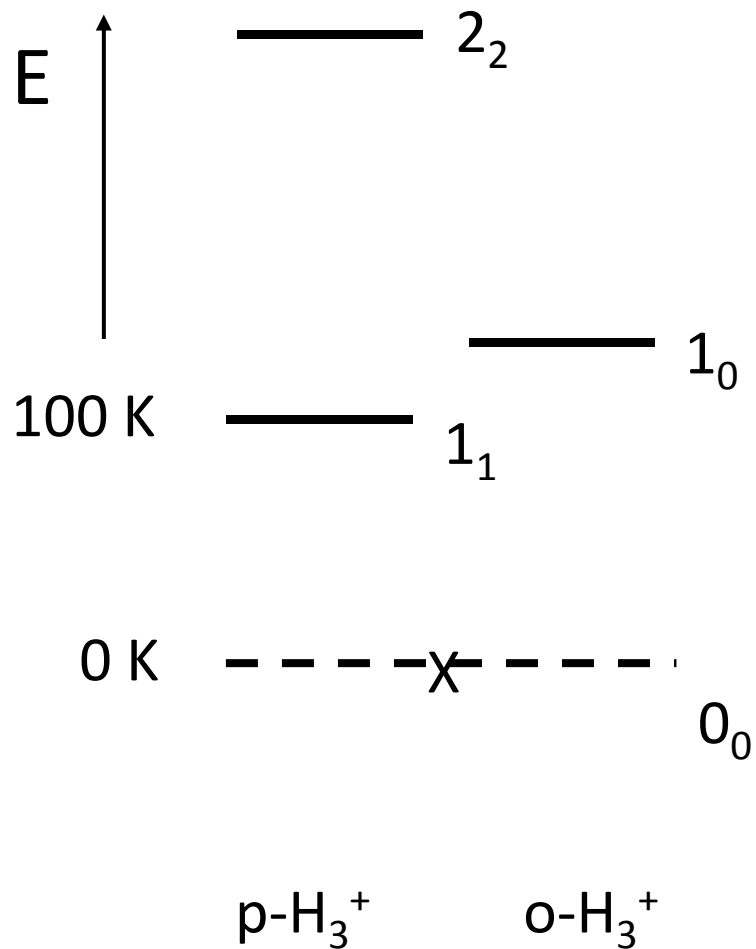
# Molecular Beam Testing



# Towards State Preparation: Collaboration with Dieter Gerlich



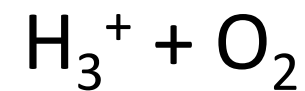
# A Chairman's Dream Experiment



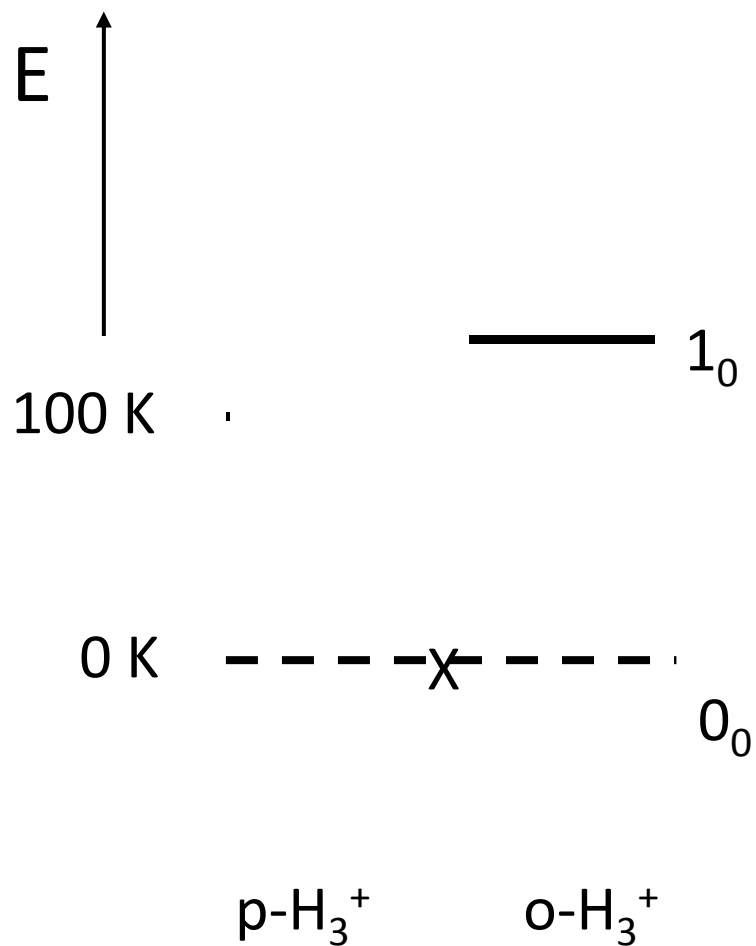
Step 1:

Prepare  $1_0$  Level

by chemical quenching



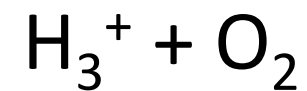
# A Chairman's Dream Experiment



Step 1:

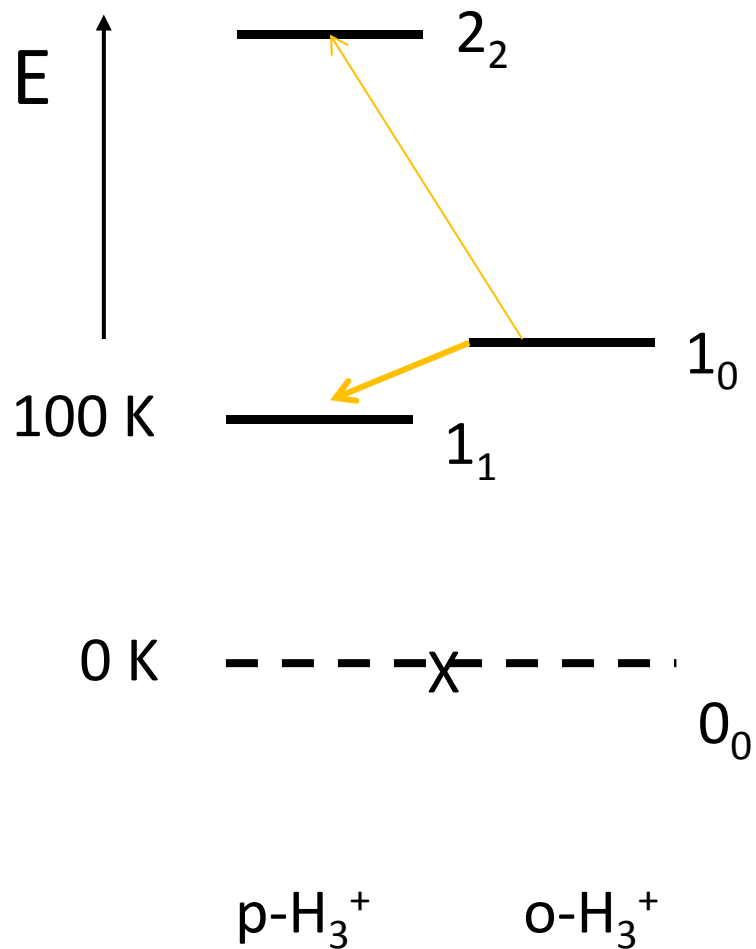
Prepare  $1_0$  Level

by chemical quenching





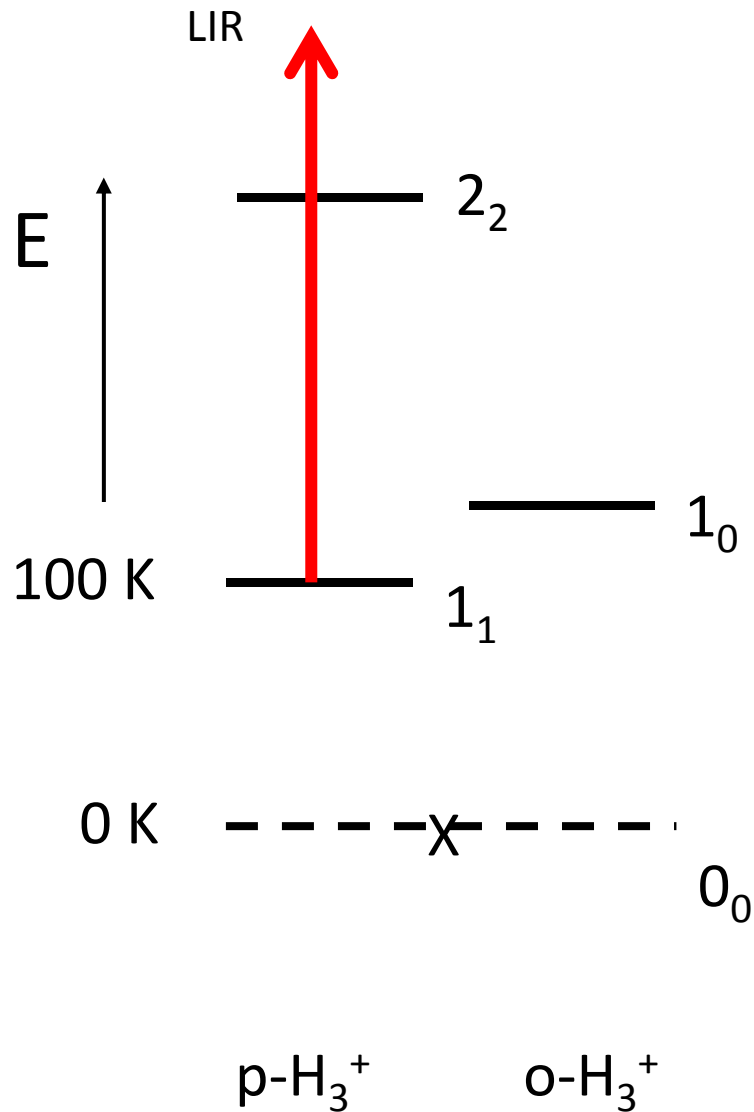
# H<sub>3</sub><sup>+</sup> Dream Experiment



Step 2:

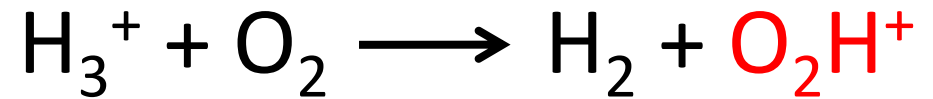
Inelastic Collisions  
e.g. with H<sub>2</sub>

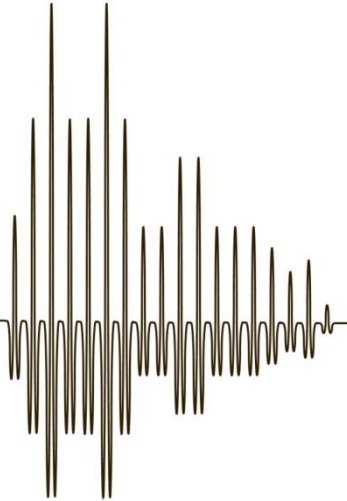
# H<sub>3</sub><sup>+</sup> Dream Experiment



Step 3:

Probe Level Population  
with LIR

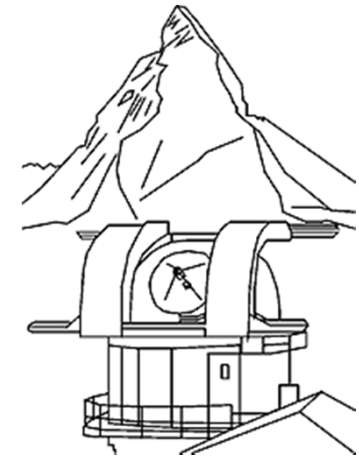




# Cold trap experiments on $\text{H}_3^+ + \text{O}_2$ proton transfer

Stephan Schlemmer

Universität zu Köln



- The THz view into the Sky: Astrophysical Picture
- $\text{H}_2\text{D}^+ / \text{H}_3^+$  in Space and Laboratory
- $\text{H}_3^+ + \text{O}_2$  Proton Transfer
- Preparing a single rotational state of  $\text{H}_3^+$